EARCH

INVENTOR SEARCH

=> d his 147

(FILE 'HCAPLUS' ENTERED AT 11:04:52 ON 30 AUG 2007)

L47 20 S L45 AND L46

=> d que 147

L16 QUE ABB=ON PLU=ON PSEUDOMONAS?

LI8 QUE ABB=ON PLU=ON COPOLYM? OR CO(W)POLYM?

L26 763 SEA FILE=HCAPLUS ABB=ON PLU=ON ("FUKUI, TATSUKI"/AU OR "HONMA, TSUTOMU"/AU OR "IMAMURA, TAKESHI"/AU OR

"KENMOKU, TAKASHI"/AU OR "KOZAKI, SHINYA"/AU OR

"MIHARA, CHIEKO"/AU OR "YANO, TETSUYA"/AU)

L27 QUE ABB=ON PLU=ON SUGAWA E?/AU

L28 786 SEA FILE=HCAPLUS ABB=ON PLU=ON L27 OR L26

L29 QUE ABB=ON PLU=ON FUKUI T?/AU L30 QUE ABB=ON PLU=ON HOMA T?/AU

L31 QUE ABB=ON PLU=ON IMAMURA T?/AU

L32 QUE ABB=ON PLU=ON KENMOKU T?/AU

L33 QUE ABB=ON PLU=ON KOZAKI S?/AU

L34 QUE ABB=ON PLU=ON MIHARA C?/AU

L35 QUE ABB=ON PLU=ON YANO T?/AU

L38 QUE ABB=ON PLU=ON (CANON(W)KABUSHIKI?)/PA,CS,SO,CO

1.41 9410 SEA FILE=HCAPLUS ABB=ON PLÚ=ON L27 OR (L29 OR L30 OR L31 OR L32 OR L33 OR L34 OR L35)

1.42 122 SEA FILE=HCAPLUS ABB=ON PLU=ON L41 AND L38

1.43 119 SEA FILE=HCAPLUS ABB=ON PLU=ON (L42 OR L28) AND L16

L45. 21 SEA FILE=HCAPLUS ABB=ON PLU=ON L43 AND L18

L46 QUE ABB=ON PLU=ON PY<2003 OR PRY<2003 OR AY<2003 OR MY<2003 OR REVIEW/DT

LA7 20 SEA FILE=HCAPLUS ABB=ON PLU=ON L45 AND L46

=> d his 161

(FILE 'MEDLINE, BIOSIS, DRUGU, EMBASE' ENTERED AT 11:14:08 ON 30 AUG 2007)

L61 7 S L60 AND L46

=> d que 161

L16 OUE ABB=ON PLU=ON PSEUDOMONAS?

L17 QUE ABB=ON PLU=ON POLYHYDRÓXYALKANOATE OR POLY(W)HYD

ROXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOATE

L26 763 SEA FILE=HCAPLUS ABB=ON PLU=ON ("FUKUI, TATSUKI"/AU OR "HONMA, TSUTOMU"/AU OR "IMAMURA, TAKESHI"/AU OR

OK HONMA, TSOTOMO /AO OK IMAMOKA, TAKESHI /AO OK "KENMOKU, TAKASHI"/AU OR "KOZAKI, SHINYA"/AU OR

"MIHARA, CHIEKO"/AU OR "YANO, TETSUYA"/AU)

L27 QUE ABB=ON PLU=ON SUGAWA E?/AU

L28 786 SEA FILE=HCAPLUS ABB=ON PLU=ON L27 OR L26

L29 QUE ABB=ON PLU=ON FUKUI T?/AU

L30 QUE ABB=ON PLU=ON HOMA T?/AU

L31 QUE ABB=ON PLU=ON IMAMURA T?/AU

L32 QUE ABB=ON PLU=ON KENMOKU T?/AU

L33 QUE ABB=ON PLU=ON KOZAKI S?/AU

L34 QUE ABB=ON PLU=ON MIHARA C?/AU

L35 QUE ABB=ON PLU=ON YANO T?/AU

L36 QUE ABB=ON PLU=ON L27 OR (L29 OR L30 OR L31 OR L32 O

R L33 OR L34 OR L35)

L38 QUE ABB=ON PLU=ON (CANON(W)KABUSHIKI?)/PA,CS,SO,CO

L46 QUE ABB=ON PLU=ON PY<2003 OR PRY<2003 OR AY<2003 OR MY<2003 OR REVIEW/DT

L57 255 SEA L28

L58 10077 SEA L36 OR L57

L59 60 SEA L58 AND L38

L60 34 SEA L59 AND (L16 OR L17)

L61 7 SEA L60 AND L46

=> dup rem [47]61

FILE HCAPLUS' ENTERED AT 11:22:00 ON 30 AUG 2007 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS) DATE

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FILE 'BIOSIS' ENTERED AT 11:22:00 ON 30 AUG 2007
Copyright (c) 2007 The Thomson Corporation
PROCESSING COMPLETED FOR L47
PROCESSING COMPLETED FOR L61
L63 27 DUP REM L47 L61 (0 DUPLICATES REMOVED)
ANSWERS '1-20' FROM FILE HCAPLUS
ANSWERS '21-27' FROM FILE BIOSIS
INVENTOR SEARCH RESULTS
```

=> d 163 1-27 ibib ed ab

L63 ANSWER 1 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 2004;430974 HCAPLUS Full-text DOCUMENT NUMBER: 141:5889

TITLE: Production of novel polyhydroxyalkanoate copolymers by Pseudomonas

INVENTOR(S): Kenmoku, Takashi; Yano,

Tetsuya; Mihara, Chieko;

Kozaki, Shinya; Honma, Tsutomu

; Fukui, Tatsuki; Imamura,

Takeshi

PATENT ASSIGNEE(S): Canon Kabushiki Kaisha,

SOURCE: Japan; Sugawa, Etsuko
PCT Int. Appl., 245 pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO.
-----WO 2004044213 A1 20040527 WO 2003-JP13531
2003
1023

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG

JP 2004162044 A 20040610 JP 2003-356748

2003 1016

AU 2003274742 A1 20040603 AU 2003-274742

2003_. 1023

US 2006211100 A1 20060921 US 2005-531689

2005 0415

PRIORITY APPLN. INFO.:

JP 2002-310250 A 2002 1024

<--

JP 2003-356748 A . 2003

2003 1016

WO 2003-JP13531 W 2003 1023

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ED Entered STN: 27 May 2004
AB
       A process is provided for the production of novel polyhydroxyalkanoate copolymers by Pseudomonas bacteria grown on various precursor fatty
       acids. Also provided is a process for the chemical oxidation of these polyhydroxyalkanoates to obtain different unique polymers.
REFERENCE COUNT:
                        13 THERE ARE 13 CITED REFERENCES AVAILABLE
                FOR THIS RECORD. ALL CITATIONS AVAILABLE
                IN THE RE FORMAT
L63 ANSWER 2 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
                          2004:370985 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                           140:357893
TITLE:
                Biosynthesis of novel polyhydroxyalkanoate
             containing 3-hydroxy-ω-(4-
             carboxyphenyl)alkanoic acid units and
             composition thereof
INVENTOR(S):
                    Fukui, Tatsuki; Yano,
             Tetsuya; Mihara, Chicko;
             Kozaki, Shinya; Honma, Tsutomu
             ; Kenmoku, Takashi
PATENT ASSIGNEE(S):
                        Canon Kabushiki Kaisha,
             Japan
SOURCE:
                  PCT Int. Appl., 278 pp.
             CODEN: PIXXD2
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                    English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                    KIND DATE
                                    APPLICATION NO.
                                                          DATE
  -----
  WO 2004037889
                          20040506 WO 2003-JP13532
                                   2003
                                   1023
    W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA,
      CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES,
      FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG,
      KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK,
      MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU,
      SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA,
      UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
    RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM,
       AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ,
       DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL,
       PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN,
      GQ, GW, ML, MR, NE, SN, TD, TG
  JP 2004315782
                   Α
                       20041111 JP 2003-356982
                                   2003
                                   1016
  AU 2003275603
                     Αl
                        20040513 AU 2003-275603
                                   2003
                                   1023
  US 2006079662
                        20060413
                                   US 2005-531226
                    A١
                                   2005
                                   0413
PRIORITY APPLN. INFO.:
                                   JP 2002-310310
                                   2002
                                   1024
                       JP 2003-92408
                                   2003
                                   0328
                       JP 2003-356982
                                   2003
                                   1016
                       WO 2003-JP13532
                                   2003
```

1023

ED Entered STN: 07 May 2004

AB A polyhydroxyalkanoate containing in a mol. thereof one or more 3-hydroxy-o-(4-carboxyphenyl)alkanoic acid units represented by chemical formula (I); wherein m is an integer selected from 0 to 7; R1 is an H, Na or K atom; and when more than one unit exists, n and R1 may differ from unit to unit, resp. Thus, 5-phenylvaleric acid and 5-(4-vinylphenyl)valeric acid were microbial copolymd, using Pseudomonas cichorii in polypeptone to a give (R)-3-Hydroxy-5-phenylvaleric acid-(R)-3-hydroxy-5-(4-vinylphenyl)valeric acid isotactic copolymer which was oxidized by MnO2 to a give (R)-3-hydroxy-5-phenylvaleric acid-(R)-3-hydroxy-5-(4- carboxyphenyl)valeric acid isotactic copolymer. The polyhydroxyalkanoate has high thermal stability, charge stability, a high charge amount, improved dispersibility and biodegradability together, and therefore, suitable for resin moldings, and binders and charge controlling agents for toners used in the electrophotog, process.

L63 ANSWER 3 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

DOCUMENT NUMBER:

2004:411682 HCAPLUS Full-text 140:414894

TITLE:

Polyhydroxyalkanoates having cyclohexyl structures in side chains, their microbial manufacture, and their use for resin binders, electrophotographic toners, and

electrophotographic image-forming process and

apparatus

INVENTOR(S):

Honma, Tsutomu; Furusaki, Shinya;

Yano, Tetsuya

PATENT ASSIGNEE(S):

Canon Inc., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 63 pp.

CODEN: JKXXAF DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.

.....

KIND DATE

APPLICATION NO.

JP 2004143288

20040520 JP 2002-309635

2002 1024

JP 3647432

B2 20050511

US 2004143087

20040722 US 2003-692206

2003

1022

PRIORITY APPLN. INFO.:

JP 2002-309635

2002

1024

OTHER SOURCE(S):

MARPAT 140:414894

ED Entered STN: 21 May 2004

Polyhydroxyalkanoates (PHA) having monomer units I (RI = H, CN, NO2, halo, Me, Et, Pr, CF3, C2F5, C3F7), useful for electrophotog, toner binders, are manufactured by culturing microorganisms in media containing alkanoic acids II (R1 = same as above). Thus, Pseudomonas cichorii YN2 (FERM BP-7375) was shake-cultured in an M9 medium containing 0.5% yeast extract and 6.0 mM 3-cyclohexylpropionic acid at 30° to give 210 mg/L of PHA comprising 94% 3-hydroxy-3-cyclohexylpropionic acid unit and 6% medium-chain-length 3-hydroxyalkanoic acid units. A magenta toner containing the PHA, C.I. Pigment Red 114, NXVP 434 (charge controlling agent), and hexamethyldisilazane-treated hydrophobic silica powder showed good charging properties and biodegradability and gave high-quality images showing no fogging even under high temperature and humidity (30°, relative humidity 80%).

L63 ANSWER 4 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION.NUMBER:

2004:57584 HCAPLUS Full-text

DOCUMENT NUMBER:

140:112164

TITLE:

Thermally stable polyhydroxyalkanoate

copolymers having ester

group-containing units on the side chain and

their manufacture

INVENTOR(S):

Imamura, Takeshi; Kenmoku,

Takashi; Fukui, Itsuki; Sugawa,

Etsuko; Yano, Tetsuya

PATENT ASSIGNEE(S):

Canon Inc., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 48 pp.

CODEN: JKXXAF DOCUMENT TYPE:

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 2004018729 20040122 JP 2002-177285 2002 0618 JP 3880462 B2 20070214 PRIORITY APPLN. INFO.: JP 2002-177285 2002 0618 Entered STN: 23 Jan 2004 AΒ The copolymers have OCH[(CH2)nCO2R]CH2CO (R = Me, Et, Me2CH, Me3C, PhCH2; n = 1-6) units and OCH[(CH2)mR']CH2CO (m = 1-8; R' = Ph, thienyl, cyclohexyl) units and are manufactured from ROCO(CH2)pCH2CO2H (p = 1-6) and R'(CH2)qCH2CO2H (q = 1-8) by biosynthesis using microorganisms. Thus, Pseudomonas cichorii YN2 was incubated in M9 medium containing polypeptone, 5-phenylvaleric acid, and monomethyl sebacate at 30° for 40 h, centrifuged, washed with MeOH, freeze-dried, extracted with CHCl3, filtered, evaporated, precipitated with cold MeOH, and dried to give a polymer having OCH[(CH2)4CO2Me]CH2CO, OCH[(CH2)2Ph]CH2CO, and OCH[(CH2)6CO2Me]CH2CO units with Mn 8.1 + 104 and Mw 15.9 + 104. L63 ANSWER 5 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 2004:57581 HCAPLUS Full-text DOCUMENT NUMBER: 140:129835 TITLE: Method for manufacture of pigmented coating compositions without the needs for dispersants INVENTOR(S): Honma, Tsutomu; Nomoto, Takeshi; Furusaki, Shinya; Yano, Tetsuya PATENT ASSIGNEE(S): Canon Inc., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 55 pp. CODEN: JKXXAF DOCUMENT TYPE: LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE -----JP 2004018723 A 20040122 JP 2002-177236 · 2002 0618 PRIORITY APPLN. INFO.: JP 2002-177236 2002. 0618 ED Entered STN: 23 Jan 2004 The method uses pigments which have surface at least partially coated or encapsulated by poly(hydroxyalkanoates) having specific pendants, graft or modification groups for improving pigment self dispersibility. The coating or encapsulation of pigments is done biol. using poly(hydroxyalkanoate) polymerase and hydroxyalkanoyl-coenzymes (as substrates). L63 ANSWER 6 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 2003:34935 HCAPLUS Full-text DOCUMENT NUMBER: 138:98160 TITLE: Microbial poly(hydroxy alkanoate)-containing charge controlling agents for use in electrostatographic developer toners in image formation method and apparatus INVENTOR(S): Yano, Tetsuya; Imamura, Takeshi; Kenmoku, Takashi; Sugawa, Etsuko PATENT ASSIGNEE(S): Canon Inc., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 43 pp. CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

DATE

KIND DATE APPLICATION NO.

PATENT NO.

```
JP 2003012778
                       20030115 JP 2001-180627
                                  2001
                                  0614
PRIORITY APPLN. INFO.:
                                  JP 2001-133552
                                  2001
                                  0427
ED Entered STN: 15 Jan 2003
AB
       The agents have monomer units shown as AmB(1-m) [A = CHRCH2CO2; B = CH((CH2)nMe)CH2CO2 and/or
       CH(CH2CH:CH(CH2)kMe)CH2CO2; m = 0.01-1; n = 0-10; k = 3, 5; R = (CH2)qC6H4R1, (CH2)rOC6H4R2, (CH2)sC6H10R3,
       (CH2)tC(O)C6H4R4; R1-R4 = H, halo, CN, NO2, CF3, C2F5, C3F7; r, q, s, t = 1-8]. Electrostatog. developer toner binders containing the agents,
       electrostatog, developer toners containing binder resins, colorants, and the agents, image formation method and apparatus using the electrostatog.
       developer toners are also claimed. The agents are biodegradable and environmentally friendly and have improved dispersibility, changeability,
       stability, etc.
1.63 ANSWER 7 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
                         2003:29442 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                          138:90972
TITLE:
                Biodegradable poly(hydroxyalkanoate) fishing
            lures and their manufacture
INVENTOR(S):
                   Sugawa, Etsuko; Imamura,
            Takeshi; Kenmoku, Takashi;
            Honma, Tsutomu; Yano, Tetsuya
PATENT ASSIGNEE(S):
                        Canon Inc., Japan
SOURCE:
                 Jpn. Kokai Tokkyo Koho, 16 pp.
            CODEN: JKXXAF
DOCUMENT TYPE:
                       Patent
LANGUAGE:
                    Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                   KIND DATE APPLICATION NO.
                                                         DATE
                       20030114 JP 2001-201112
  JP 2003009721
                                  2001
                                  0702
PRIORITY APPLN. INFO.:
                                  JP 2001-201112
                                  2001
                                  0702
ED Entered STN: 14 Jan 2003
       AB
       cyclohexyl, COPh, 2-thienyl, 2-thienylcarbonyl; Y = (CH2)nMe, CH2CH:CH(CH2)kMe; b = 1-8; n = 0-10; k = 3, 5; m = 0.01-1]. Thus,
       Pseudomonas cichorii YN2 FERM BP-7375 was cultured in a medium containing 5-phenylvaleric acid to give polymer comprising 96% 3-
       hydroxy-5-phenylbutyrate unit and 4% 3-hydroxybutyrate unit, which was molded to give a rod-like test piece showing good flexibility, recovery
       from bending, tensile strength, and biodegradability.
L63 ANSWER 8 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
                         2003:693179 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                          139:214919
TITLE:
               Polyhydroxyalkanoates containing a vinylphenyl
            structure in their side chain and method of
            manufacturing the same
INVENTOR(S):
                    Honma, Tsutomu; Sugawa,
             Etsuko; Yano, Tetsuya;
            Imamura, Takeshi; Kenmoku,
            Takashi; Fukui, Tatsuki
PATENT ASSIGNEE(S): Canon Kabushiki Kaisha,
            Japan
                 Eur. Pat. Appl., 56 pp.
SOURCE:
            CODEN: EPXXDW
DOCUMENT TYPE:
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PATENT NO. KIND DATE APPLICATION NO. DATE

English

LANGUAGE:

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

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EP 1340778
                       20030903 EP 2003-4350
                                    0228
  EP 1340778
                       20060118
                   B1
     R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,
       MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ,
       EE, HU, SK
  JP 2004238592
                        20040826 JP 2003-37322
                                    0214
  JP 3689700
                  B2 20050831
  US 2003204044
                     A1
                         20031030
                                    US 2003-372285
                                    2003
                                    0225
  US 6645743
                       20031111
                   B2
  CN 1445257
                                   CN 2003-106688
                       20031001
                                    2003
                                    0228
PRIORITY APPLN. INFO.:
                                    JP 2002-54897
                                    2002
                                    0228
                       JP 2002-362962
                                    2002
                                    1213
                        JP 2003-37322
                                    2003
                                    0214
ED Entered STN: 05 Sep 2003
       A polyhydroxyalkanoate (PHA) having a desired configuration is produced using a raw material containing ω-(4-vinylphenyl)- alkanoic acid and
       w-substituted alkanoic acid in which a group having a ring structure selected from Ph, thienyl, and cyclohexyl structures substitutes therefor on the
       end thereof by producing a PHA copolymer containing the corresponding 3-hydroxy-ω-(4-vinylphenyl)-alkanoate unit and the corresponding 3-
       hydroxy-ω-substituted alkanoate unit by making use of a microorganism capable of producing the PHA or by oxidizing a predetd, portion of the
       corresponding PHA. Pseudomonas cichorii YN2 was cultivated with 5-(4-vinylphenyl)-valeric acid and 5-phenyl-valeric acid to give a
       polyhydroxyalkanoate.
REFERENCE COUNT:
                             THERE ARE 6 CITED REFERENCES AVAILABLE
                FOR THIS RECORD. ALL CITATIONS AVAILABLE
                IN THE RE FORMAT
L63 ANSWER 9 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
                          2003:653269 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                           139:180526
TITLE:
                Polyhydroxyalkanoate copolymer
             including unit having bromo group in side
             chain and production method thereof
INVENTOR(S):
                     Honma, Tsutomu; Kozaki,
             Shinya; Imamura, Takeshi;
             Kenmoku, Takashi; Fukui,
             Tatsuki; Sugawa, Etsuko;
             Yano, Tetsuya
PATENT ASSIGNEE(S):
                         Canon Kabushiki Kaisha,
             Japan
                  Eur. Pat. Appl., 46 pp.
SOURCE:
             CODEN: EPXXDW
DOCUMENT TYPE:
LANGUAGE:
                     English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                 , KIND DATE
                                     APPLICATION NO.
                                                            DATE
  EP 1336634
                   A2 20030820 EP 2003-3418
                                    2003
                                    0214
                         <--.
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EP 1336634

20040107

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EP 1336634
                    B1 20051123
     R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,
       MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ,
       EE, HU, SK
  JP 2004196832
                         20040715 JP 2002-362594
                                     2002
                                     1213
                          <--
  JP 3754956
                   B2 20060315
   US 2003194789
                      A1 20031016 US 2003-359600
                                     2003
                                     0207
  US 7135540
                    B2
                         20061114
  CN 1438258
                         20030827
                                    CN 2003-104451
                                     2003
                                     0214
PRIORITY APPLN. INFO.:
                                      JP 2002-39255
                                     2002
                                     0215
                          ۷..
                         JP 2002-310268
                                          Α
                                     2002
                                     1024
                         JP 2002-362594
                                     2002
                                     1213
ED Entered STN: 22 Aug 2003
AB
        The title polyhydroxyalkanoate copolymer, prepared by microorganisms using a ω-bromoalkanoic acid, is thermally stable and capable of
        arbitrarily\ controlling\ phys.\ properties.\ The\ polyhydroxyalkanoate\ copolymer\ includes\ a\ 3-hydroxy-\omega-bromoalkanoic\ acid\ unit\ of\ arbitrarily\ controlling\ phys.
        OCH[(CH2)nBr|CH2CO (n = 1-8) and a unit of OCH[(CH2)mR]CH2CO (m = 1-8; R = Ph, thienyl, cyclohexyl) within a same mol.
L63 ANSWER 10 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER:
                           2002:944749 HCAPLUS Full-text
DOCUMENT NUMBER:
                            138:14534
TITLE:
                 Biodegradable cards comprising
              polyhydroxyalkanoates with good extrudability,
              heat resistance, and mechanical properties
INVENTOR(S):
                      Yano, Tetsuya; Imamura,
              Takeshi; Kenmoku, Takashi;
              Sugawa, Etsuko
PATENT ASSIGNEE(S): Canon Inc., Japan
SOURCE:
                   Jpn. Kokai Tokkyo Koho, 23 pp.
              CODEN: JKXXAF.
DOCUMENT TYPE:
                         Patent
LANGUAGE:
                      Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                     KIND DATE
                                       APPLICATION NO.
                                                              DATE
  JP 2002356542
                         20021213 JP 2001-164772
                                     2001
                                     0531
PRIORITY APPLN, INFO.:
                                      JP 2001-164772
                                     2001
                                     0531
ED Entered STN: 13 Dec 2002
        The cards, useful for credit cards, IC cards, etc., contain polyhydroxyalkanoates having a repeating unit of OCHXCH2CO [X = (CH2)aR1,
        (CH2)bCH2C6H4R2, (CH2)cOC6H4R3, (CH2)dCH2C6H10R4, (CH2)eCOC6H4R5, (CH2)fCH2Y, (CH2)gCOY; R1 = H (a = 5,10), halo (a = 1-
        10), chromophore (a = 1-10), carboxy (a = 1-10), oxiranyl (a = 1-7); R2-4 = H, halo, CN, NO2, CF3, C2F5, C3F7; R5 = same as R2-4, Me, Et, Pr;
        Y = thienyl; b, d, f = 0-7; c, e, g = 1-8]. The cards may have layers for printing, thermal recording, magnetic recording, and embossing. Thus, a
        multilayer card comprising D-glucose-4-fluorobenzenepentanoic acid copolymer [prepared in culture media containing Pseudomonas cichorii YN
        2 (FERM BP-7375)] showed tensile strength 24.0 MPa and good embossability.
```

ACCESSION NUMBER:

2002:847920 HCAPLUS Full-text

DOCUMENT NUMBER:

137:338648

TITLE:

Polyhydroxyalkanoate/thermoplastic resin blend

compositions, their biodegradable heat-resistant moldings, and their

thermoforming

INVENTOR(S): Honma, Tsutomu; Imamura.

Takeshi; Kenmoku, Takashi; Sugawa, Etsuko; Yano, Tetsuya Canon Inc., Japan

PATENT ASSIGNEE(S):

Jpn. Kokai Tokkyo Koho, 10 pp.

SOURCE: CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 2002322355 20021108 JP 2001-131691

> 2001 0427

PRIORITY APPLN. INFO.:

JP 2001-131691

2001

0427

ED Entered STN: 08 Nov 2002

AΒ The compns., useful for containers for foods, beverages, cosmetics, and pharmaceutics, etc., contain (A) polyhydroxyalkanoates containing 1-99 mol% 3-hydroxy-5-benzoylvaleric acid (1)-derived monomer units OCH(CH2COPh)CH2CO and (B) thermoplastic resins such as polyesters, polystyrenes, polypropylenes, poly(ethylene terephthalates), polyvinyls, and polyamides. Thus, a dry composition of 81:10:5:3:1 1-3hydroxydecanoic acid-3-hydroxyoctanoic acid-3-hydroxyhexanoic acid-3- hydroxydodecanoic acid copolymer, prepared with Pseudomonas cichorii H45 from 5-benzoylvaleric acid was mixed with styrenic polymer (Styron 685) at ratio 50:50, sheeted by foam extrusion, and secondary molded into bowls for instant noodles having good biodegradability, high hardness and low brittleness at 25 and 100°, resp., Tg 40°, and Tm 156°.

L63 ANSWER 12 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2002:847533 HCAPLUS Full-text

DOCUMENT NUMBER:

137:353523

TITLE:

Microbial production of polyhydroxyalkanoates

for charge control agents for toners

INVENTOR(S): Imamura, Takeshi; Sugawa,

Etsuko; Yano, Tetusya;

Kenmoko, Takashi

PATENT ASSIGNEE(S): Canon Kabushiki Kaisha,

Japan

SOURCE:

Eur. Pat. Appl., 82 pp. CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

EP 1254918

20021106 EP 2002-9696

2002

0429

EP 1254918 A3 20030226 EP 1254918 BI 20040331

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,

MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2003048968 20030221 JP 2002-125613

2002

0426

<--

JP 3848204 B2 20061122

US 2003073804 20030417 US 2002-133379 **A**1

2002

0429

US 6855472 B2 20050215

PRIORITY APPLN. INFO.:

JP 2001-131693 2001

0427

۷.. JP 2001-131811 2001

0427

ED Entered STN: 08 Nov 2002

AB Polyhydroxyalkanoates are prepared from thienylsulfanyl alkanoic acids by microbial polymerization. The polyhydroxyalkanoates are useful as biodegradable charge control agents having excellent charging characteristics, excellent dispersibility in the toner resin and improved spent property. 5-(2-Thienylsulfanyl) valeric acid was cultured with Pseudomonas cichorii YN2 to give a polyhydroxyalkanoate.

L63 ANSWER 13 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2002:831899 HCAPLUS Full-text

DOCUMENT NUMBER:

137:343851

TITLE:

Electrostatic charge image developing toner

and image forming method

INVENTOR(S):

Yano, Tetsuya; Nomoto, Tsuyoshi;

Kozaki, Shinya; Honma, Tsutomu

PATENT ASSIGNEE(S):

Canon Kabushiki Kaisha,

Japan

SOURCE:

Eur. Pat. Appl., 80 pp. CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

English FAMILY ACC. NUM. COUNT: 3

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

EP 1253475

A2 20021030 EP 2002-9673

2002 0429

EP 1253475

20031126 A3

EP 1253475 BI 20070808

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,

MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2003015359 A 20030117 JP 2001-210021

2001

0710

PRIORITY APPLN. INFO.:

JP 2001-133728

2001

0427

JP 2001-210021 Α

> 2001 0710

OTHER SOURCE(S):

MARPAT 137:343851

ED · Entered STN: 01 Nov 2002

Electrostatic charge image developing toner allows to design the toner characteristics such as chargeability, flowability, stability in time and environmental stability uniform among the toners of different colors. The toner has a small particle size enough for enabling uniform dispersion and being excellent in color saturation and transparency. The toner also shows higher contribution to the environmental security. The toner includes a coloring agent of which at least a part of the surface is covered with polyhydroxyalkanoate (PHA). The toner is produced by dispersing the coloring agent in aqueous medium, then fixing PHA synthesizing enzyme to the coloring agent dispersed in the aqueous medium, then adding 3-hydroxyacyl CoA, and executing a PHA synthesizing reaction to cover at least a part of the surface of the coloring agent with PHA. The toner thus obtained is used for an image forming method.

L63 ANSWER 14 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2002:831842 HCAPLUS Full-text

DOCUMENT NUMBER:

137:331172

TITLE:

Manufacture of biodegradable polyhydroxyalkanoates having phenylsulfinyl and/or phenylsulfonyl terminal structure on side chains and their use as charge control agents in electrophotographic toner binder for image forming apparatus

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10/531,689
INVENTOR(S):
                     Imamura, Takeshi; Sugawa,
              Etsuko; Yano, Tetsuya;
             Kenmoku, Takashi
PATENT ASSIGNEE(S): Canon Kabushiki Kaisha,
             Japan
SOURCE:
                  Eur. Pat. Appl., 99 pp.
             CODEN: EPXXDW
DOCUMENT TYPE:
LANGUAGE:
                     English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                    KIND DATE
                                     APPLICATION NO.
                                                           DATE
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  .....
  EP 1253162
                   A2 20021030 EP 2002-9675
                                    2002
                                    0429
  EP 1253162
                   A3 20030226
  EP 1253162
                   ΒI
                       20060419
    R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,
       MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
  JP 2003034716
                        20030207
                                   JP 2002-127700
                    Α
                                    2002
                                   0426
  JP 3880444
                  B2 20070214
  US 2003100700
                         20030529
                                    US 2002-133671
                                    2002
                                    0429
                         <--
  US 6808854
                   B2 20041026
PRIORITY APPLN. INFO.:
                                    JP 2001-131831
                                    2001
                                    0427
                        JP 2001-133640
                                    2001
                                    0427
ED Entered STN: 01 Nov 2002
AB
        The present invention provides a novel polyhydroxyalkanoate (PHA) containing a 3-hydroxyalkanoic unit which has at its side chain terminal a
        substituted phenylsulfinyl group and/or a substituted phenylsulfonyl group, and a production process thereof. The novel PHA can be produced by
        oxidizing with a peroxide a biosynthetic PHA containing a 3-hydroxyalkanoic unit which has at its side chain terminal a substituted phenylsulfanyl
        group. The novel PHA has a superior function as a charge control agent, besides is biodegradable, hence is contributable to environmental
        conservation. Thus, a typical polymer was prepared by transforming 5-(phenylsulfanyl)valeric acid in a culture of pseudomonas cichorii YN2,
        followed by oxidizing the polyhydroxyalkanoate with H2O2.
L63 ANSWER 15 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2002:831841 HCAPLUS Full-text
DOCUMENT NUMBER:
                           137:326082
TITLE:
                 Microbial preparation of polyhydroxyalkanoates
             for charge control agents and binders for
             toners
INVENTOR(S):
                     Kenmoku, Takashi; Kobayashi, Toyoko;
             Sugawa, Etsuko; Yano, Tetsuya
             ; Kobayashi, Shin; Imamura, Takeshi;
              Honma, Tsutomu
PATENT ASSIGNEE(S):
                         Canon Kabushiki Kaisha,
             Japan
SOURCE:
                  Eur. Pat. Appl., 83 pp.
             CODEN: EPXXDW
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                     English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                    KIND DATE
                                     APPLICATION NO.
                                                           DATE
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A2 20021030 EP 2002-9667

2002

EP 1253161

0429

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EP 1253161
                  20030226
               A3
EP 1253161
               ВI
                   20060322
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R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,

MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2003034717 20030207 Α JP 2002-127588

2002 0426

<--

B2 20061122 US 2003104300 A1 20030605 US 2002-133576

2002

0429

US 6908720 B2 20050621

PRIORITY APPLN. INFO.:

JP 3848206

JP 2001-133651

2001 0427

JP 2001-133667 Α

2001 0427

ED Entered STN: 01 Nov 2002

AB The present invention provides a polyhydroxyalkanoate (PHA) having a unit containing thioether with high reactivity, and its production method. The present invention also provides a charge control agent containing the PHA; a toner binder containing the charge control agent; an electrostatic latent image developing toner containing the charge control agent; and an image forming method and an image forming apparatus using the electrostatic latent image developing toner.

L63 ANSWER 16 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2002:831840 HCAPLUS Full-text

DOCUMENT NUMBER:

137:338415

TITLE:

Preparation of microbial polyhydroxyalkanoates

and their use in toners

INVENTOR(S):

Honma, Tsutomo; Yano, Tetsuya;

Nomoto, Tsuyoshi; Kozaki, Shinya

PATENT ASSIGNEE(S): Canon Kabushiki Kaisha,

Japan

SOURCE:

Eur. Pat. Appl., 60 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE: English FAMILY ACC, NUM, COUNT: 3

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

EP 1253160

A2 20021030 EP 2002-9695

2002 0429

EP 1253160 A3 20031022

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,

MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR

JP 2003011312 20030115 JP 2001-208704 2001

0710

<--

B2 20050817

JP 3684175 PRIORITY APPLN. INFO.:

JP 2001-131694

2001

0427

JP 2001-208704

2001 0710

Entered STN: 01 Nov 2002

A construct comprises a base material and a polyhydroxyalkanoate, wherein at least a part of the base material is coated with the AB polyhyroxyalkanoate, and the polyhydroxyalkanoate comprises a 3-hydroxyalkanoic acid unit other than 3-hydroxypropionic acid unit, 3-hydroxyn-butyric acid unit, and 3-hydroxy-n-valeric acid unit. A method for making a construct comprises: immobilizing a medium chain length

polyhydroxyalkanoate synthetase to a base material, and reacting 3-hydroxyacyl CoA with the synthetase to synthesize a polyhydroxyalkanoate and to coat at least a part of the base material with the polyhdroxyalkanoate. Polyhydroxyalkanoate synthetase was isolated from a transformant having polyhydroxyalkanoate synthetase production capacity, immobilized on alumina, and incubated with (R)-3-hydroxyoctanoyl CoA to give a polyhydroxyalkanoate.

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L63 ANSWER 17 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN
                          2002:752333 HCAPLUS Full-text
ACCESSION NUMBER:
DOCUMENT NUMBER:
                           137:263769
TITLE:
                 Thienyl-containing polyhydroxyalkanoates
             prepared by microbial polymerization, and use
             in charge control agents, toner binders and
             image-forming apparatus
INVENTOR(S):
                     Yano, Tetsuya; Sugawa,
             Etsuko; Imamura, Takeshi;
             Honma, Tsutomu; Kenmoku,
             Takashi
PATENT ASSIGNEE(S):
                         Canon Kabushiki Kaisha,
             Japan
SOURCE:
                   Eur. Pat. Appl., 83 pp.
             CODEN: EPXXDW
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                     English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                    KIND DATE
                                     APPLICATION NO.
                                                            DATE
  EP 1245605
                       20021002 EP 2002-7025
                                    2002
                                    0327
  EP 1245605
                       20030402
  EP 1245605
                   BL
                       20070314
     R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,
       MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
  US 2003096182
                     A1 20030522 US 2002-105305
                                    2002
                                    0326
                         <--
  US 6777153
                        20040817
  JP 2003012786
                        20030115
                                   JP 2002-89658
                                    2002
                                    0327
  JP 3745298
                  B2 20060215
PRIORITY APPLN. INFO.:
                                    JP 2001-90026
                                    2001
                                    0327
                       JP 2001-133551
                                    2001
                                    0427
ED Entered STN: 04 Oct 2002
AB
       The title polyhydroxyalkanoate comprise units represented by OCHRCH2CO, where R is (CH2)nCOR1, R1 is 2-thienyl, and n is 1 to 8. Also
       disclosed are a process for producing the polyhydroxyalkanoate by the use of a microorganism having the ability to produce the
       polyhydroxyalkanoate and accumulate it in the bacterial body; a charge control agent, a toner binder and a toner which contain this
       polyhydroxyalkanoate; and an image-forming method and an image-forming apparatus which make use of the toner.
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L63 ANSWER 18 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2002:671943 HCAPLUS Full-text

DOCUMENT NUMBER:

137:224074

TITLE:

Novel polyhydroxyalkanoate containing unit phenylsulfanyl structure in the side chain, process for its production, charge control agent, toner binder and toner which contain novel polyhydroxyalkanoate, and image-forming apparatus which make use of the toner

INVENTOR(S):

): Imamura, Takeshi; Sugawa, Etsuko; Yano, Tetsuya; Nomoto, Tsuyoshi; Suzuki, Tomohiro; Honma,

```
Tsutomu; Kenmoku, Takashi;
             Fukui, Tatsuki
PATENT ASSIGNEE(S):
                        Canon Kabushiki Kaisha,
             Japan
SOURCE:
                  Eur. Pat. Appl., 133 pp.
             CODEN: EPXXDW
DOCUMENT TYPE:
                       Patent
LANGUAGE:
                    English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
  PATENT NO.
                   KIND DATE
                                    APPLICATION NO.
                                                         DATE
  EP 1236755
                      20020904
                                EP 2002-4759
                                  2002
                        <--
  EP 1236755
                  A3 20030402
    R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,
      MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
  US 2003013841
                    A1 20030116 US 2002-84172
                                  2002
                                  0228
                        <--
  US 7045321
                  B2 20060516
  JP 2003306534
                   Α
                       20031031
                                  JP 2002-56654
                                  2002
                                  0301
                        <-- ·
  JP 3592306
                 B2 20041124
  US 2005250191
                        20051110
                                  US 2005-155599
                                  2005
                                  0620
PRIORITY APPLN. INFO.:
                                   JP 2001-57142
                                  2001
                                  0301
                        <--
                      JP 2001-57145
                                  2001
                                  0301
                        ۷--
                      JP 2001-164774
                                  2001
                                  0531
                        <--
                      JP 2001-210037
                                       Λ
                                  2001
                                  0710
                        <--
                      JP 2001-210049
                                  2001
                                  0710
                        <--
                      JP 2002-39254
                                      Α
                                  2002
                                  0215
```

ED Entered STN: 06 Sep 2002

<--US 2002-84172

72 A3 2002 0228

AB A polyhydroxyalkanoate is disclosed which has, in the mol., a unit represented by OCH{(CH2)xSC6H4R]CH2CO, wherein R is arbitrarily selected from a hydrogen atom, a halogen atom, CN, NO2, COOR', SO2R" CH3, C2H5, C3H7, C(CH3)2H and C(CH3)3; where R' is H, Na, K, CH3 or C2H5, and R" is OH, ONa, OK, a halogen atom, OCH3 or OC2H5; and x is an integer arbitrarily selected from 1 to 8; with the proviso that a polyhydroxyalkanoate is excluded which has a hydrogen atom as R and x in all the units is 2 or 4. Also disclosed is a process for producing the polyhydroxyalkanoate by the use of a microorganism having the ability to produce the polyhydroxyalkanoate and accumulate it in the bacterial body. This polyhydroxyalkanoate is useful as a biodegradable charge-control agent in electrophotog, toner binders. A typical polymer was manufactured by inoculating 200 mL Na2HPO4 6.2, K2HPO4 3.0, NaCl 0.5, and NH4Cl 1 g/L containing 0.5% polypeptone, 0.1% 5-(phenylthio)valeric acid with Pseudomonas cichorii strain YN2, and shaking 30 h.

L63 ANSWER 19 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 2001:892025 HCAPLUS Full-text DOCUMENT NUMBER: 136:19159 TITLE: Separation of poly(hydroxyalkanoic acids) from microbial cells INVENTOR(S): Imamura, Takeshi; Kenmoku, Takashi; Suzuki, Tomohiro; Honma, Tsutomu; Nomoto, Takeshi; Sugawa, Etsuko; Yano, Tetsuya PATENT ASSIGNEE(S): Canon Inc., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. APPLICATION NO. DATE >. KIND DATE JP 2001340095 JP 2001-67254

20011211 2001

0309

PRIORITY APPLN. INFO.:

JP 2000-88715

2000 0328

ED Entered STN: 11 Dec 2001

AB The polymers are separated from cells by microwave irradiation Ralstonia eutropha TB 64 was cultured in a medium containing Na pyruvate to give cells containing poly(3-hydroxybutyric acid) (I), which was treated in a microwave oven, freeze-dried, extracted with CHCl3, and washed with McOH to give I with 87.1% purity and 85.9% recovery.

L63 ANSWER 20 OF 27 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: DOCUMENT NUMBER: 2000:616368 HCAPLUS Full-text 133:206898

TITLE:

Core-shell or multilayered polymer particles

and their in situ manufacture with

microorganisms

INVENTOR(S):

Kawahata, Yuji; Imamura, Takeshi

PATENT ASSIGNEE(S): Canon Inc., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 13 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 2000236892 20000905. JP 1999-39955 1999

PRIORITY APPLN. INFO.:

JP 1999-39955

1999 0218

0218

ED Entered STN: 06 Sep 2000

The particles, which might be useful for sustained-release pharmaceuticals, are manufactured by cultivation of microorganisms capable of producing polymer particles in the cells, in media containing materials for the polymers with changing the concns. and/or components of the materials. Burkholderia cepacia KK01 strain (FERM BP-4235) was cultured in a medium containing Na pyruvate, centrifuged, cultured in another medium containing Na valerate, centrifuged to collect the microorganism, and treated with NaOCI to manufacture perfectly spherical particles consisted of poly(3-hydroxybutyric acid) as the core and poly(3-hydroxyvaleric acid) as the shell.

L63 ANSWER 21 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2003:68616 BIOSIS Full-text

DOCUMENT NUMBER: PREV200300068616

TITLE: Method for producing microbial polyester. AUTHOR(S): Imamura, Takeshi [Inventor, Reprint

Author]; Yano, Tetsuya [Inventor];

Kobayashi, Shin [Inventor]; Suda, Sakae [Inventor];

Honma, Tsutomu [Inventor]

CORPORATE SOURCE: Chigasaki, Japan

ASSIGNEE: Canon Kabushiki

Kaisha, Tokyo, Japan

PATENT INFORMATION: US 6492147 20021210

SOURCE:

Official Gazette of the United States Patent and

Trademark Office Patents, (Dec 10 2002)

Vol. 1265, No. 2. http://www.uspto.gov/web/menu/pat

data.html, e-file.

ISSN: 0098-1133 (ISSN print).

DOCUMENT TYPE:

Patent English

LANGUAGE: ENTRY DATE:

Entered STN: 29 Jan 2003

Last Updated on STN: 29 Jan 2003

ED Entered STN: 29 Jan 2003

Last Updated on STN: 29 Jan 2003

AΒ A method for producing a microbial polyester by culturing a microorganism being capable of producing a poly hydroxyalkanoate polyester in a culture medium containing 1-hexene as a sole carbon source.

L63 ANSWER 22 OF 27 BIOSIS COPYRIGHT. (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2003:55194 BIOSIS Full-text

DOCUMENT NUMBER: PREV200300055194

TITLE:

Polyhydroxyalkanoate synthase and gene

encoding the same enzyme.

AUTHOR(S):

Yano, Tetsuya [Inventor, Reprint Author];

Imamura, Takeshi [Inventor]; Suda, Sakae

[Inventor]; Honma, Tsutomu [Inventor]

CORPORATE SOURCE: Atsugi, Japan

ASSIGNEE: Canon Kabushiki

Kaisha, Tokyo, Japan

PATENT INFORMATION: US 6485951 20021126

SOURCE:

Official Gazette of the United States Patent and

Trademark Office Patents, (Nov 26 2002)

Vol. 1264, No. 4. http://www.uspto.gov/web/menu/pat

data.html. e-file.

ISSN: 0098-1133 (ISSN print).

DOCUMENT TYPE: Patent

LANGUAGE: **English**

ENTRY DATE: Entered STN: 22 Jan 2003

Last Updated on STN: 22 Jan 2003

ED Entered STN: 22 Jan 2003

Last Updated on STN: 22 Jan 2003

A novel polyhydroxyalkanoate (PHA) synthase derived from a microorganism capable of producing a PHA having a novel side-chain structure and a DNA encoding the amino acid sequence for the synthase are provided. Two PHA synthase proteins (SEQ ID Nos. 1 and 3) derived from Pseudomonas jessenii P161 (FERM BP-7376) and PHA synthase genes encoding these PHA synthases are provided, respectively (SEQ ID Nos. 2 and 4). A recombinant microorganism is endowed with a PHA producing ability.

L63 ANSWER 23 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2003:53351 BIOSIS Full-text

DOCUMENT NUMBER: PREV200300053351

TITLE:

Polyhydroxyalkanoate containing

3-hydroxythienylalkanoic acid as monomer unit and

method for producing the same.

Honma, Tsutomu [Inventor, Reprint AUTHOR(S):

Author]; Yano, Tetsuya [Inventor];

Kobayashi, Shin [Inventor]; Imamura.

Takeshi [Inventor]; Kenmoku, Takashi

[Inventor]; Kozaki, Shinya [Inventor]

CORPORATE SOURCE: Atsugi, Japan

ASSIGNEE: Canon Kabushiki

Kaisha, Tokyo, Japan

PATENT INFORMATION: US 6479621 20021112

SOURCE: Official Gazette of the United States Patent and

Trademark Office Patents, (Nov 12 2002)

Vol. 1264, No. 2. http://www.uspto.gov/web/menu/pat

data.html. e-file.

ISSN: 0098-1133 (ISSN print).

DOCUMENT TYPE: Patent

LANGUAGE:

English

ENTRY DATE:

Entered STN: 22 Jan 2003

Last Updated on STN: 22 Jan 2003

ED Entered STN: 22 Jan 2003

Last Updated on STN: 22 Jan 2003

Microorganisms capable of synthesizing novel polyhydroxyalkanoate having 3-hydroxythienylalkanoic acid as monomer unit, using thienylalkanoic acid as a stock are cultured on a culture medium containing thienylalkanoic acid, and the polyhydroxyalkanoate produced in the cultured cell is extracted and

L63 ANSWER 24 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2001:192930 BIOSIS Full-text

DOCUMENT NUMBER: PREV200100192930

TITLE:

Pseudomonas cepacia strain isolated from termite intestines that degrades trichlorethylene

and furan compounds.

AUTHOR(S): Kato, Kiny [Inventor, Reprint author]; Kozaki,

Shinya [Inventor]; Imamura, Takeshi [Inventor]; Sakuranaga, Masanori [Inventor]

CORPORATE SOURCE: Atsugi, Japan

ASSIGNEE: Canon Kabushiki-

Kaisha, Tokyo, Japan

PATENT INFORMATION: US 6096530 20000801

SOURCE:

Official Gazette of the United States Patent and

Trademark Office Patents, (Aug. 1, 2000)

Vol. 1237, No. 1. e-file.

CODEN: OGUPE7. ISSN: 0098-1133.

DOCUMENT TYPE: Patent LANGUAGE: English

ENTRY DATE: Entered STN: 20 Apr 2001

Last Updated on STN: 18 Feb 2002

ED Entered STN: 20 Apr 2001

Last Updated on STN: 18 Feb 2002

A biologically pure culture of Pseudomonas cepacia strain KK01 (FERM BP-4235) is capable of degrading trichloroethylene. A method for obtaining microorganisms having a trichloroethylene degrading ability comprises the steps of culturing microorganisms separated from the bodies of termites in a culture medium. A method for remediating a soil contaminated with trichloroethylene comprises the steps of providing a soil contaminated with trichloroethylene and bringing microorganisms having a trichloroethylene degrading ability derived from intestine of termites into contact with the soil, and biodegrading trichloroethylene in the soil. A method for biodegrading trichloroethylene, comprises the steps of culturing Pseudomonas cepacia KK01 (FERM BP-4235) under existence of an inducer and inducing Pseudomonas cepacia KK01 (FERM BP-4235) to have ability for degrading trichloroethylene and bringing Pseudomonus cepacia KK01 having trichloroethylene degrading ability to contact with trichloroethylene and biodegrading trichloroethylene.

L63 ANSWER 25 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2000:331469 BIOSIS Full-text

DOCUMENT NUMBER: PREV200000331469

TITLE: Remedying a contaminated environment using

Pseudomonas cepacia or Corynebacterium species and Renobacter species FERM BP-5353 having

dehalogenase activity.

AUTHOR(S): Imamura, Takeshi [Inventor, Reprint

author]; Yano, Tetsuy [Inventor] CORPORATE SOURCE: Chigasaki, Japan

ASSIGNEE: Canon Kabushiki

Kaisha, Tokyo, Japan

PATENT INFORMATION: US 6017746 20000125

SOURCE:

Official Gazette of the United States Patent and

Trademark Office Patents, (Jan. 25, 2000)

Vol. 1230, No. 4. e-file.

CODEN: OGUPE7. ISSN: 0098-1133.

DOCUMENT TYPE: Patent LANGUAGE: **English**

ENTRY DATE: Entered STN: 2 Aug 2000

Last Updated on STN: 7 Jan 2002

ED Entered STN: 2 Aug 2000

Last Updated on STN: 7 Jan 2002

A process for remedying an environment contaminated with an aliphatic organochlorine compound which includes the use of Pseudomouas cepacia strain KK01 (FERM BP-4235) or Corynebacterium species (FERM BP 5102) and Renobacter species (FERM BP-5353). The first two microorganisms are capable of introducing an oxygen atom into the aliphatic organochlorine compound in order to convert the aliphatic compound to an epoxide. During protonization the epoxide is converted into a chlorinated organic acid. Renobacter species strain FERM BP-5353 decomposes chlorinated organic acids to substances naturally existing in nature. The chlorinated and/or halogenated acids include chloroacetic

acid, dichloroacetic acid, trichloroacetic acid and dichloropropionic acid, etc. The polluted environments in which the processes may be carried out include the soil, ground water and waste water.

L63 ANSWER 26 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2002:83352 BIOSIS Full-text

DOCUMENT NUMBER: PREV200200083352

TITLE: Bacterial KB2.

AUTHOR(S): Imamura, T. [Inventor]; Yano, T.

[Inventor]

CORPORATE SOURCE: Chigasaki, Japan

ASSIGNEE: CANON KABUSHIKI

KAISHA

PATENT INFORMATION: US 5665597 19970909

SOURCE: Official Gazette of the United States Patent and

Trademark Office Patents, (Sept. 9, 1997) Vol. 1202, No. 2, pp. 1328. print. CODEN: OGUPE7. ISSN: 0098-1133.

DOCUMENT TYPE: Patent

LANGUAGE: English

ENTRY DATE: Entered STN: 16 Jan 2002

Last Updated on STN: 25 Feb 2002

ED Entered STN: 16 Jan 2002

Last Updated on STN: 25 Feb 2002

L63 ANSWER 27 OF 27 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2002:82147 BIOSIS Full-text

DOCUMENT NUMBER: PREV200200082147

TITLE: Method for biodegradation of polluting substance.

AUTHOR(S): Kato, K. [Inventor]; Tanaka, K. [Inventor];

Sakuranaga, M. [Inventor]; Kozaki, S.

[Inventor]

CORPORATE SOURCE: Yokohama, Japan

ASSIGNEE: CANON KABUSHIKI

KAISHA

PATENT INFORMATION: US 5658795 19970819

SOURCE: Official Gazette of the United States Patent and

Trademark Office Patents, (Aug. 19, 1997)

Vol. 1201, No. 3, pp. 2099. print.

CODEN: OGUPE7. ISSN: 0098-1133.

DOCUMENT TYPE: Patent

LANGUAGE: English

ENTRY DATE: Entered STN: 16 Jan 2002

Last Updated on STN: 25 Feb 2002

ED Entered STN: 16 Jan 2002

Last Updated on STN: 25 Feb 2002

STRUCTURE SEARCH

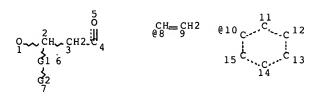
=> d his 152

(FILE 'HICAPLUS' ENTERED AT 11:04:52 ON 30 AUG 2007) 11 S L50 NOT L47

=> d que stat 152 1.4

147 L48

1.49



REP G1=(1-8) CH2 VAR G2=8/10 NODE ATTRIBUTES: CONNECT IS E1 RC AT 5 DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 15

STEREO ATTRIBUTES: NONE 1299 SEA FILE=REGISTRY SSS FUL L4 1.6 L12 949 SEA FILE=HCAPLUS ABB=ON PLU=ON L6 57652 SEA FILE=HCAPLUS ABB=ON PLU=ON PSEUDOMONAS+PFT,OLD,NT L14 91 SEA FILE=HCAPLUS ABB=ON PLU=ON L12 AND L14 L15 L16 QUE ABB=ON PLU=ON PSEUDOMONAS? QUE ABB=ON PLU=ON POLYHYDROXYALKANOATE OR POLY(W)HYD L17 ROXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOATE QUE ABB=ON PLU=ON COPOLYM? OR CO(W)POLYM? L18 QUE ABB=ON PLU=ON L17(3A)L18 L19: 99 SEA FILE=HCAPLUS ABB=ON PLU=ON L12 AND L16 1.20 99 SEA FILE=HCAPLUS ABB=ON PLU=ON L15 OR L20 1.21 2313 SEA FILE=HCAPLUS ABB=ON PLU=ON POLYHYDROXYALKANOATE 1.22 OR POLY(W)HYDROXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOAT E L23 78 SEA FILE=HCAPLUS ABB=ON PLU=ON L21 AND L22 1.24 8 SEA FILE=HCAPLUS ABB=ON PLU=ON L21 AND L19 1.26 763 SEA FILE=HCAPLUS ABB=ON PLU=ON ("FUKUI, TATSUKI"/AU OR "HONMA, TSUTOMU"/AU OR "IMAMURA, TAKESHI"/AU OR "KENMOKU, TAKASHI"/AU OR "KOZAKI, SHINYA"/AU OR "MIHARA, CHIEKO"/AU OR "YANO, TETSUYA"/AU) QUE ABB=ON PLU=ON SUGAWA E?/AU L27 L28 786 SEA FILE=HCAPLUS ABB=ON PLU=ON L27 OR L26 L29 QUE ABB=ON PLU=ON FUKULT?/AU QUE ABB=ON PLU=ON HOMA T?/AU 1.30 QUE ABB=ON PLU=ON IMAMURA T?/AU 1.31 L32 QUE ABB=ON PLU=ON KENMOKU T?/AU L33 QUE ABB=ON PLU=ON KOZAKI S?/AU L34 QUE ABB=ON PLU=ON MIHARA C?/AU L35 QUE ABB=ON PLU=ON YANO T?/AU QUE ABB=ON PLU=ON (CANON(W)KABUSHIKI?)/PA,CS,SO,CO L38 1.41 9410 SEA FILE=HCAPLUS ABB=ON PLU=ON L27 OR (L29 OR L30 OR L31 OR L32 OR L33 OR L34 OR L35) 122 SEA FILE=HCAPLUS ABB=ON PLU=ON 1.41 AND L38 IA21.43 119 SEA FILE=HCAPLUS ABB=ON PLU=ON (L42 OR L28) AND L16 21 SEA FILE=HCAPLUS ABB=ON PLU=ON LA3 AND L18 L45 QUE ABB=ON PLU=ON PY<2003 OR PRY<2003 OR AY<2003 OR L46 MY<2003 OR REVIEW/DT 20 SEA FILE=HCAPLUS ABB=ON PLU=ON 1.45 AND 1.46

24 SEA FILE=HCAPLUS ABB=ON PLU=ON L23 AND L18 24 SEA FILE=HCAPLUS ABB=ON PLU=ON L48 OR L24

L50 21 SEA FILE=HCAPLUS ABB=ON PLU=ON L49 AND L46 L52 11 SEA FILE=HCAPLUS ABB=ON PLU=ON L50 NOT L47

=> d his 162

(FILE 'MEDLINE, BIOSIS, DRUGU, EMBASE' ENTERED AT 11:14:08 ON 30 AUG 2007)

L62 4 S L56 NOT L61

=> d que stat 162 L4 STR

0 1 61 61 62 CH=CH2 010 C C 12
15 C C 13

REP G1=(1-8) CH2
VAR G2=8/10
NODE ATTRIBUTES:
CONNECT IS E1 RC AT 5
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 15

STEREO ATTRIBUTES: NONE

- L6 1299 SEA FILE=REGISTRY SSS FUL L4
- L10 20 SEA FILE=REGISTRY ABB=ON PLU=ON L6 AND (MEDLINE/LC

OR BIOSIS/LC OR DRUGU/LC OR EMBASE/LC)

- L16 QUE ABB=ON PLU=ON PSEUDOMONAS?
- L17 QUE ABB=ON PLU=ON POLYHYDROXYALKANOATE OR POLY(W)HYD

ROXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOATE

L26 763 SEA FILE=HCAPLUS ABB=ON PLU=ON ("FUKUI, TATSUKI"/AU OR "HONMA, TSUTOMU"/AU OR "IMAMURA, TAKESHI"/AU OR

"KENMOKU, TAKASHI"/AU OR "KOZAKI, SHINYA"/AU OR

"MIHARA, CHIEKO"/AU OR "YANO, TETSUYA"/AU)

- L27 QUE ABB=ON PLU=ON SUGAWA E?/AU
- L28 786 SEA FILE=HCAPLUS ABB=ON PLU=ON L27 OR L26
- 1.29 QUE ABB=ON PLU=ON FUKUI T?/AU
- L30 QUE ABB=ON PLU=ON HOMA T?/AU
- L31 QUE ABB=ON PLU=ON IMAMURA T?/AU
- L32 QUE ABB=ON PLU=ON KENMOKU T?/AU
- 1.33 QUE ABB=ON PLU=ON KOZAKI S?/AU
- L34 QUE ABB=ON PLU=ON MIHARA C?/AU
- L35 QUE ABB=ON PLU=ON YANO T?/AU
- L36 QUE ABB=ON PLU=ON L27 OR (L29 OR L30 OR L31 OR L32 O

R L33 OR L34 OR L35)

- L38 QUE ABB=ON PLU=ON (CANON(W)KABUSHIKI?)/PA,CS,SO,CO
- L46 QUE ABB=ON PLU=ON PY<2003 OR PRY<2003 OR AY<2003 OR
 - MY<2003 OR REVIEW/DT
- L53 67 SEA L10
- L54 3 SEA L53 AND L16
- L55 2 SEA L53 AND L17
- £56 4 SEA £54 OR £55
- L57 255 SEA L28
- L58 10077 SEA L36 OR L57
- L59 60 SEA L58 AND L38
- L60 34 SEA L59 AND (L16 OR L17)
- L61 7 SEA L60 AND L46
- L62 4 SEA L56 NOT L61

=> dup rem 152 162

FILE 'HCAPLUS' ENTERED AT 11:23:35 ON 30 AUG 2007

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FILE 'MEDLINE' ENTERED AT 11:23:35 ON 30 AUG 2007

FILE 'BIOSIS' ENTERED AT 41:23:35 ON 30 AUG 2007 Copyright (c) 2007 The Thomson Corporation PROCESSING COMPLETED FOR L52 PROCESSING COMPLETED FOR L62 L64 15 DUP REM L52 L62 (0 DUPLICATES REMOVED)

ANSWERS '1-11' FROM FILE HCAPLUS ANSWER '12' FROM FILE MEDLINE ANSWERS '13-15' FROM FILE BIOSIS

STRUCTURE SEARCH RESULTS

=> d 164 1-11 ibib ed abs hitstr hitind

L64 ANSWER I OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: DOCUMENT NUMBER:

2001:912506 HCAPLUS Full-text 136:200571

TITLE:

Preparation and Characterization of Enantiomerically Pure Telechelic Diols from

mcl-Poly[(R)-3-hydroxyalkanoates]

AUTHOR(S):

Andrade, Austin P.; Witholt, Bernard; Hany,

CORPORATE SOURCE:

Roland; Egli, Thomas; Li, Zhi Institute of Biotechnology, ETH-Zuerich,

Honggerberg, Zurich, CH-8093, Switz.

SOURCE:

Macromolecules (2002), 35(3),

684-689

CODEN: MAMOBX; ISSN: 0024-9297

PUBLISHER: American Chemical Society DOCUMENT TYPE:

LANGUAGE:

Journal

English ED Entered STN: 19 Dec 2001

Novel enantiomerically pure telechelic OH-terminated poly[(R)-3-hydroxyoctanoate] (PHO-diol), poly[(R)-3- hydroxyoctanoate-co-poly[(R)-3-ΑB hydroxy-7-oxooctanoate] (PHOO-diol), and poly[(R)-3-hydroxyoctanoate-co-poly[(R)-3-hydroxy-7-octenoate] (PHUO-diol) have been synthesized in 80-91% yield from the corresponding high mol. weight polymers, resp., by catalytic transesterification with ethylene glycol. The number average mol. wts. (Mn) of these telechelic diols reached (2.0-3.0) + 103, which corresponds to 17-20 repeated monomer units. For PHOOdiol and PHUO-diol, the side chain functional groups remained, which provides with addnl. reactive groups for further polymerization or modification. The structures of the diols were confirmed by 1H NMR and IR spectra. The glass transition temps. (Tg) of the telechelic diols are between -46 and -56 °C and the melting transition temps. (Tm) are lower than 40 °C, all determined by DSC. These telechelic diols can be used as soft-segments to prepare novel block copolymers with desired properties.

IT 401495-09-8, (R)-3-Hydroxyhexanoate-(R)-3-hydroxyoctanoate-

(R)-3-Hydroxy-5-hexenoate-(R)-3-hydroxy-7-octenoate

copolymer, ester with ethylene glycol (1:1)

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation and characterization of enantiomerically pure telechelic diols from medium chain length-poly[(R)-3-

hydroxyalkanoates])

RN 401495-09-8 HCAPLUS

CN 7-Octenoic acid, 3-hydroxy-, (3R)-, polymer with

(3R)-3-hydroxyhexanoic acid, (3R)-3-hydroxy-5-hexenoic acid and (3R)-3-hydroxyoctanoic acid, 2-hydroxyethyl ester (9CI) (CA INDEX NAME)

CM I

CRN 107-21-1 CMF C2 H6 O2

HO_CH2_CH2_OH

CM₂

CRN 128971-78-8

CMF (C8 H16 O3 . C8 H14 O3 . C6 H12 O3 . C6 H10 O3)x

CCI PMS

CM₃

CRN 119003-50-8 CMF C8 H14 O3

Absolute stereochemistry.

CM 4

CRN 119003-49-5 CMF C6 H10 O3

Absolute stereochemistry.

CM 5

CRN 77877-35-1 CMF C6 H12 O3

Absolute stereochemistry. Rotation (-).

CM 6

CRN 44987-72-6 CMF C8 H16 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 $(CH_2)_4$
 Me

CC 35-8 (Chemistry of Synthetic High Polymers)

ST transesterification ethylene glycol telechelic enantiomerically pure polyhydroxyalkanoate diol synthesis

IT Pseudomonas putida

(GPo1; preparation and characterization of enantiomerically pure telechelic diols from medium chain length-poly[(R)-3-hydroxyalkanoates])

1T 107-21-1, Ethylene glycol, reactions 401495-04-3,

(R)-3-Hydroxynexanoate-(R)-3-hydroxyoctanoate copolymer, ester with ethylene glycol (1:1) 401495-06-5,

(R)-3-Hydroxyhexanoate-(R)-3-hydroxyoctanoate-(R)-3-Hydroxy-5-oxohexanoate-(R)-3-hydroxy-7-oxooctanoate copolymer,

ester with ethylene glycol (1:1) 401495-09-8,

(R)-3-Hydroxyhexanoate-(R)-3-hydroxyoctanoate-(R)-3-Hydroxy-5-

hexenoate-(R)-3-hydroxy-7-octenoate copolymer, ester with ethylene glycol (1:1)

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation and characterization of enantiomerically pure telechelic diols from medium chain length-poly[(R)-3-

hydroxyalkanoates])

REFERENCE COUNT: 53 THERE ARE 53 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L64 ANSWER 2 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2001:921063 HCAPLUS Full-text

DOCUMENT NUMBER:

136:231306

TITLE:

Genetically engineered Pseudomonas:

A factory of new bioplastics with broad

applications

AUTHOR(S):

Olivera, Elias R.; Carnicero, David; Jodra,

Ruth; Minambres, Baltasar; Garcia, Belen; Abraham, Gustavo A.; Gallardo, Alberto; San Roman, Julio; Garcia, Jose L.; Naharro,

German; Luengo, Jose M.

CORPORATE SOURCE:

Departamento de Bioquimiça y Biologia

Molecular, Facultad de Veterinaria, Universidad de Leon, Leon, 24007, Spain

SOURCE:

3(10), 612-618

Environmental Microbiology (2001),

CODEN: ENMIFM; ISSN: 1462-2912

PUBLISHER: DOCUMENT TYPE: Blackwell Science Ltd.

Journal

LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 136:231306

ED Entered STN: 21 Dec 2001

New bioplastics containing aromatic or mixts. of aliphatic and aromatic monomers have been obtained using genetically engineered strains of Pseudomonas putida. The mutation (-) or deletion (Δ) of some of the genes involved in the β-oxidation pathway (fadA-, fadB- ΔfadA or ΔfadBA mutants) elicits a strong intracellular accumulation of unusual homo-or co-polymers that dramatically alter the morphol. of these bacteria, as more than 90% of the cytoplasm is occupied by these macromols. The introduction of a blockade in the β -oxidation pathway, or in other related catabolic routes, has allowed the synthesis of polymers other than those accumulated in the wild type (with regard to both monomer size and relative percentage), the accumulation of certain intermediates that are rapidly catabolized in the wild type and the accumulation in the culture broths of end metabolites that, as in the case of phenylacetic acid, phenylbutyric acid, trans-cinnamic acid or their derivs., have important medical or pharmaceutical applications (antitumoral, analgesic, radiopotentiators, chemopreventive or antihelmintic). Furthermore, using one of these polyesters (poly 3-hydroxy-6-phenylhexanoate), we obtained polymeric microspheres that could be used as drug vehicles.

IT 247169-46-6P 402956-11-0P 402956-12-1P

402956-13-2P 402956-14-3P 402956-15-4P

402956-16-5P

RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic preparation); PRP (Properties); BIOL (Biological study); PREP

(bioplastic production with genetically engineered

Pseudomonas)

RN 247169-46-6 HCAPLUS

CN Benzenehexanoic acid, β-hydroxy-, (βR)-, homopolymer (9CI) (CA INDEX NAME)

CM I

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

RN 402956-11-0 HCAPLUS

CN Benzeneheptanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenepentanoic acid, isotactic (9CI) (CA INDEX NAME)

CM 1

CRN 247169-47-7 CMF C13 H18 O3

Absolute stereochemistry.

CM 2

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

$$HO_2C$$

RN 402956-12-1 HCAPLUS

CN. Benzeneheptanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenehexanoic acid and (βR)-β-hydroxybenzenepentanoic acid, isotactic (9Cl) (CA INDEX NAME)

CM 1

CRN 247169-47-7 CMF C13 H18 O3

Absolute stereochemistry.

CM 2

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

. CM 3

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 Ph

RN 402956-13-2 HCAPLUS

CN Benzeneoctanoic acid, β-hydroxy-, (βR)-, polymer with

 $(\beta R)\text{-}\beta\text{-hydroxybenzenehexanoic acid, isotactic (9CI)}$ (CA INDEX NAME)

CM 1

CRN 247169-51-3 CMF C14 H20 O3

Absolute stereochemistry.

CM₂

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

RN 402956-14-3 HCAPLUS

CN Benzeneoctanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzeneheptanoic acid, (βR)-β-hydroxybenzenehexanoic acid and (βR)-β-hydroxybenzenepentanoic acid, isotactic (9CI) (CA INDEX NAME)

CM I

CRN 247169-51-3 CMF C14 H20 O3

Absolute stereochemistry.

CM 2.

CRN 247169-47-7 CMF C13 H18 O3

Absolute stereochemistry.

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

CM 4

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

RN 402956-15-4 HCAPLUS

CN Benzeneoctanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenehexanoic acid, (3R)-3-hydroxyhexanoic acid and (3R)-3-hydroxyoctanoic acid, isotactic (9CI) (CA INDEX NAME)

CM 1

CRN 247169-51-3 CMF C14 H20 O3

Absolute stereochemistry.

CM 2

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

CM 3

CRN 77877-35-1 CMF C6 H12 O3

Absolute stereochemistry. Rotation (-).

CM 4

CRN 44987-72-6 CMF C8 H16 O3

Absolute stereochemistry.

RN 402956-16-5 HCAPLUS

CN Poly[oxy[(3R)-1-oxo-3-(3-phenylpropyl)-1,3-propanediyl]] (9Cl) (CA INDEX NAME)

IT 247169-45-5

RL: BSU (Biological study, unclassified); BIOL (Biological study) (bioplastic production with genetically engineered Pseudomonas)

RN 247169-45-5 HCAPLUS

CN Benzenehexanoic acid, β-hydroxy-, (βR)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

CC 16-4 (Fermentation and Bioindustrial Chemistry)

ST Pseudomonas fermn polyhydroxyalkanoate prodn

IT Fermentation

Genetic engineering.

Glass transition temperature

Microspheres

Polydispersity

Pseudomonas putida

(bioplastic production with genetically engineered

Pseudomonas)

IT Gene, microbial

RL: BSU (Biological study, unclassified); BIOL (Biological study)

(fadB; bioplastic production with genetically engineered

Pseudomonas)

IT Gene, microbial

RL: BSU (Biological study, unclassified); BIOL (Biological study) (fadD; bioplastic production with genetically engineered

Pseudomonas)

```
IT Polyesters, preparation
    RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic
    preparation); PRP (Properties); BIOL (Biological study); PREP
     (hydroxycarboxylic acid-based; bioplastic production with
     genetically engineered Pseudomonas)
 IT Mutagenesis
     (transposon; bioplastic production with genetically engineered
     Pseudomonas)
 IT 124-07-2, Octanoic acid, processes 142-62-1, Hexanoic acid.
    processes 156-38-7, 4-Hydroxy-phenylacetic acid 2270-20-4,
    5-Phenylvaleric acid 5581-75-9, 6-Phenylhexanoic acid
    26547-51-3, 8-Phenyloctanoic acid 40228-90-8, 7-Phenylheptanoic
    RL: BCP (Biochemical process); BIOL (Biological study); PROC
    (Process)
     (bioplastic production with genetically engineered
     Pseudomonas)
 IT 247169-46-6P 402956-11-0P 402956-12-1P
    402956-13-2P 402956-14-3P 402956-15-4P
    402956-16-5P
    RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic
    preparation); PRP (Properties); BIOL (Biological study); PREP
    (Preparation)
      (bioplastic production with genetically engineered
     Pseudomonas)
 IT 247169-45-5
    RL: BSU (Biological study, unclassified); BIOL (Biological study)
     (bioplastic production with genetically engineered
     Pseudomonas)
REFERENCE COUNT:
                            35 THERE ARE 35 CITED REFERENCES AVAILABLE
                  FOR THIS RECORD. ALL CITATIONS AVAILABLE
                   IN THE RE FORMAT
 L64 ANSWER 3 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER:
                             2001:180517 HCAPLUS Full-text
 DOCUMENT NUMBER:
                              134:367300
 TITLE:
                   Microbial Synthesis of Poly(β-
               hydroxyalkanoates) Bearing Phenyl Groups from
               Pseudomonas putida: Chemical Structure
               and Characterization
 AUTHOR(S):
                       Abraham, Gustavo A.; Gallardo, Alberto; San
               Roman, Julio; Olivera, Elias R.; Jodra, Ruth;
               Garcia, Belen; Minambres, Baltasar; Garcia,
               Jose L.; Luengo, Jose M.
 CORPORATE SOURCE:
                             Instituto de Ciencia y Tecnologia de
               Polimeros, CSIC, Madrid, 28006, Spain
 SOURCE:
                     Biomacromolecules (2001), 2(2),
               562-567
               CODEN: BOMAF6; ISSN: 1525-7797
 PUBLISHER:
                       American Chemical Society
 DOCUMENT TYPE:
                           Journal
 LANGUAGE:
                        English
 ED Entered STN: 16 Mar 2001
         New poly(β-hydroxyalkanoates) having aroms, groups (so-called PHPhAs) from a microbial origin have been characterized. These polymers were
         produced and accumulated as reserve materials when a $\beta$-oxidation mutant of Pseudomonas putida U, disrupted in the gene that encodes the 3-
         ketoacyl-CoA thiolase (fadA), was cultured in a chemical defined medium containing different aromatic fatty acids (6-phenylhexanoic acid, 7-
         phenylheptanoic acid, a mixture of them, or 8-phenyloctanoic acid) as carbon sources. The polymers were extracted from the bacteria, purified and
         characterized by using 13C NMR spectroscopy, gel permeation chromatog., and differential scanning calorimetry. Structural studies revealed that
         when 6-phenylhexanoic acid was added to the cultures, an homopolymer (poly-3-hydroxy-6-phenylhexanoate) was accumulated. The feeding with
         8-phenyloctanoic acid and 7-phenylheptanoic acid leads to the formation of copolymers of the corresponding units with the n - 2 carbons formed
         after deacetylation, copoly(3-hydroxy-8- phenyloctanoate-3-hydroxy-6-phenylhexanoate) and copoly(3-hydroxy-7-phenylheptanoate-3-hydroxy-5-
         phenylvalerate), resp. The mixture of 6-phenylhexanoic acid and 7-phenylheptanoic acid gave rise to the corresponding terpolymer, copoly(3-
         hydroxy-7-phenylheptanoate-3-hydroxy-6-phenylhexanoate-3- hydroxy-5-phenylvalerate). Studies on the chemical structure of these three
         polyesters revealed that they were true copolymers but not a mixture of homopolymers and that the different monomeric units were randomly
          incorporated in the macromol, chains. Thermal behavior and mol, weight distribution were also discussed. These compds, had a dual attractive
```

interest in function of (i) their broad use as biodegradable polymers and (ii) their possible biomedical applications. IT 172923-04-5P 247169-52-4P 340255-66-5P

340255-71-2P 340255-73-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(microbial preparation of polyesters containing Ph group with

Pseudomonas putida)

RN 172923-04-5 HCAPLUS

CN Benzenepentanoic acid, β -hydroxy-, (βR)-, homopolymer,

isotactic (9CI) (CA INDEX NAME)

CM 1

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

RN 247169-52-4 HCAPLUS

CN Benzeneoctanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenehexanoic acid (9CI) (CA INDEX NAME)

CM I

CRN 247169-51-3 . CMF C14 H20 O3

Absolute stereochemistry.

CM₂

CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 $(CH_2)_3$
 Ph

RN 340255-66-5 HCAPLUS

CN Poly[oxy[(3R)-1-oxo-3-(2-phenylethyl)-1,3-propanediyl]] (9Cl) (CA INDEX NAME)

RN 340255-71-2 HCAPLUS

CN Benzeneheptanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenepentanoic acid (9Cl) (CA INDEX NAME)

CM I

CRN 247169-47-7

CMF C13 H18 O3

Absolute stereochemistry.

CM 2

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 Ph

RN 340255-73-4 HCAPLUS

CN Benzeneheptanoic acid, β-hydroxy-, (βR)-, polymer with (βR)-β-hydroxybenzenehexanoic acid and (βR)-β-hydroxybenzenepentanoic acid (9CI) (CA INDEX NAME)

CM 1 . .

CRN 247169-47-7 CMF C13 H18 O3

Absolute stereochemistry.

CM 2

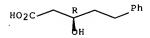
CRN 247169-45-5 CMF C12 H16 O3

Absolute stereochemistry.

CM 3

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.



CC 35-5 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 10

ST microbial polyhydroxyalkanoate phenyl contg prepn characterization; Pseudomonas putida polyester prepn characterization; phenylhexanoic acid microbial polyester prepn; phenylheptanoic acid microbial polyester prepn; phenyloctanoic acid microbial polyester prepn

IT Pseudomonas putida

(in preparation of polyesters containing Ph group)

Polyesters, preparation

RL: SPN (Synthetic preparation); PREP (Preparation)

(microbial preparation of polyesters containing Ph group with

Pseudomonas putida)

IT 172923-04-5P 247169-52-4P 340255-66-5P

340255-71-2P 340255-73-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(microbial preparation of polyesters containing Ph group with

Pseudomonas putida)

REFERENCE COUNT:

29 THERE ARE 29 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L64 ANSWER 4 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN 2001:856979 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER:

136:196742

TITLE:

Intracellular degradation of two structurally

different polyhydroxyalkanoic acids accumulated in Pseudomonas putida and Pseudomonas citronellolis from mixtures of octanoic acid and 5-phenylvaleric

AUTHOR(S):

Chung, Dong Min; Choi, Mun Hwan; Song, Jae Jun; Yoon, Sung Chul; Kang, Inn-Kyu; Huh, Nam

Eung

CORPORATE SOURCE:

Biomaterials Science Laboratory, Division of

Applied Life Sciences at The Graduate School, Gyeongsang National University, Jinju,

660-701, S. Korea

SOURCE:

International, Journal of Biological

Macromolecules (2001), 29(4-5),

243-250

CODEN: IJBMDR; ISSN: 0141-8130

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: LANGUAGE:

Journal English

ED Entered STN: 27 Nov 2001

From a set of mixed carbon sources, 5-phenylvaleric acid (PV) and octanoic acid (OA), polyhydroxyalkanoic acid (PHA) was sep. accumulated in the two pseudomonads Pseudomonas putida BM01 and Pseudomonas citronellolis (ATCC 13674) to investigate any structural difference between the two PHA accumulated under a similar culture condition using one-step culture technique. The resulting polymers were isolated by chloroform solvent extraction and characterized by fractional precipitation and differential scanning calorimetry. The solvent fractionation anal, showed that the PHA synthesized by P. putida was separated into two fractions, 3-hydroxy-5-phenylvalerate (3HPV)-rich PHA fraction in the precipitate phase and 3-hydroxyoctanoate (3HO)-rich PHA fraction in the solution phase whereas the PHA produced by P. citronellolis exhibited a rather little compositional separation into the two phases. According to the thermal anal., the P. putida PHA exhibited two glass transitions indicative of the PHA not being homogeneous whereas the P. citronellolis PHA exhibited only one glass transition. It was found that the structural heterogeneity of the P. putida PHA was caused by a significant difference in the assimilation rate between PV and OA. The structural heterogeneity present in the P. putida PHA was also confirmed by a first order degradation kinetics anal, of the PHA in the cells. The two different first-order degradation rate consts. (k1), 0.087 and 0.015/h for 3HO- and 3HPV-unit, resp., were observed in a polymer system over the first 20 h of degradation. In the later degradation period, the disappearance rate of 3HO-unit was calculated to be 0.020 h. The k1 value of 0.083/h, almost the same as for the 3HO-unit in the P. putida PHA, was obtained for the P(3HO) accumulated in P. putida BM01 grown on OA as the only carbon source. In addition, the k1 value of 0.015/h for the 3HPV-unit in the P. putida PHA, was also close to 0.019/h for the P(3HPV) homopolymer accumulated in P. putida BM01 grown on PV plus butyric acid. On the contrary, the k1 values for the P. citronellolis PHA were determined to be 0.035 and 0.029/h for 3HO- and 3HPV-unit, resp., thus these two relatively close values implying a random copolymer nature of the P. citronellolis PHA. In addition, the faster degradation of P(3HO) than P(3HPV) by the intracellular P. putida PHA depolymerase indicates that the enzyme is more specific against the aliphatic PHA than the aromatic PHA.

IT 129645-03-0P 134736-36-0P 401612-76-8P

RL: BSU (Biological study, unclassified); PEP (Physical,

engineering or chemical process); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); PREP (Preparation); PROC (Process)
 (intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in Pseudomonas putida and P. citronellolis from mixts. of octanoic acid and phenylvaleric acid)

RN 129645-03-0 HCAPLUS

CN Poly[oxy[1-oxo-3-(2-phenylethyl)-1,3-propanediyl]] (9Cl) (CA INDEX NAME)

RN 134736-36-0 HCAPLUS

CN Benzenepentanoic acid, β-hydroxy-, homopolymer (9CI) (CA INDEX NAME)

CM I

CRN 41479-99-6 CMF C11 H14 O3

RN 401612-76-8 HCAPLUS

CN Benzenepentanoic acid, β -hydroxy-, polymer with 3-hydroxyoctanoic acid (CA INDEX NAME)

CM I

CRN 41479-99-6 CMF C11 H14 O3

CM 2

CRN 14292-27-4 CMF C8 H16 O3

CC 10-2 (Microbial, Algal, and Fungal Biochemistry)
ST polyhydroxyalkanoate accumulation structure biodegrdn
Pseudomonas; octanoate phenylvalerate
polyhydroxyalkanoate accumulation Pseudomonas
IT Decomposition kinetics

(biodegrdn.; intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in

Pseudomonas putida and P. citronellolis from mixts. of octanoic acid and phenylvaleric acid)

IT Carbon sources, microbial

Glass transition

Pseudomonas citronellolis

Pseudomonas putida

(intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in Pseudomonas putida and P. citronellolis from mixts. of octanoic acid and phenylvaleric acid)

IT Polyesters, biological studies

RL: BSU (Biological study, unclassified); PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); PREP (Preparation); PROC (Process)

(intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in Pseudontonas putida and P. citronellolis from mixts. of octanoic acid and phenylvaleric acid)

1T 124-07-2, Octanoic acid, biological studies 2270-20-4,
 5-Phenylvaleric acid 140208-16-8, Polyhydroxyalkanoate depolymerase

RL: BSU (Biological study, unclassified); BIOL (Biological study) (intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in Pseudomonas putida and P. citronellolis from mixts. of octanoic acid and phenylvaleric acid)

TT 86175-71-5P 120659-38-3P, Poly 3 hydroxyoctanoic acid 129645-03-0P 134736-36-0P 401612-76-8P

RL: BSU (Biological study, unclassified); PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); PREP (Preparation); PROC (Process)

(intracellular degradation of two structurally different polyhydroxyalkanoic acids accumulated in Pseudomonas putida and P. citronellolis from mixts, of octanoic acid and phenylvaleric acid)

REFERENCE COUNT:

JNT: 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L64 ANSWER 5 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

2000:309693 HCAPLUS Full-text

DOCUMENT NUMBER: TITLE: PhaG-1

MBER: 133:57635 PhaG-mediated synthesis of

poly(3-hydroxyalkanoates) consisting of medium-chain-length constituents from nonrelated carbon sources in recombinant

Pseudomonas fragi

AUTHOR(S):

Fiedler, Silke; Steinbuchel, Alexander; Rehm,

Bernd H. A.

CORPORATE SOURCE: Institut fur Mikrobiologie, Westfalische

Wilhelms-Universitat Munster, Munster,

D-48149, Germany

SOURCE:

Applied and Environmental Microbiology (

2000), 66(5), 2117-2124

CODEN: AEMIDF; ISSN: 0099-2240

PUBLISHER:

American Society for Microbiology

DOCUMENT TYPE: Journal LANGUAGE: English ED Entered STN: 14 May 2000

Recently, a new metabolic link between fatty acid de novo biosynthesis and biosynthesis of poly(3-hydroxyalkanoate) consisting of medium-chain-length constituents (C6 to C14) (PHAMCL), catalyzed by the 3-hydroxydecanoyl-[acyl-carrier- protein]:CoA transacylase (PhaG), has been identified in Pseudomonas putida. To establish this PHA-biosynthetic pathway in a non-PHA-accumulating bacterium, phaC1 (encoding PHA synthase 1) from Pseudomonas aeruginosa and phaG (encoding the transacylase) from P. putida were functionally coexpressed in Pseudomonas fragi. The recombinant strains of P. fragi were cultivated on gluconate as the sole carbon source, and PHA accumulation to about 14% of the total cellular dry weight was achieved. The resp. polyester was isolated, and GPC anal. revealed a weight average molar mass of about 130,000 g mol-1 and a polydispersity of 2.2. The PHA was composed mainly (60 mol%) of 3-hydroxydecanoate. These data strongly suggested that functional expression of phaC1 and phaG established a new pathway for PHAMCL biosynthesis from nonrelated carbon sources in P. fragi. When fatty acids were used as the carbon source, no PHA accumulation was observed in PHA synthase-expressing P. fragi, whereas application of the β-oxidation inhibitor acrylic acid mediated PHAMCL accumulation. The substrate for the PHA synthase PhaC1 is therefore presumably directly provided through the enzymic activity of the transacylase PhaG by the conversion of (R)-3-hydroxydecanoyl-ACP to (R)-3-hydroxydecanoyl-CoA when the organism is cultivated on gluconate. Here we demonstrate for the first time the establishment of PHAMCL synthesis from nonrelated carbon

sources in a non-PHA-accumulating bacterium, employing fatty acid de novo biosynthesis and the enzymes PhaG (a transacylase) and PhaC1 (a PHA synthase).

IT 278185-67-4DP, copolymer with

polyhydroxyalkanoates

RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic preparation); BSU (Biological study, unclassified); MFM (Metabolic formation); BIOL (Biological study); FORM (Formation, nonpreparative); PREP (Preparation)

(PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

RN 278185-67-4 HCAPLUS

CN 11-Dodecenoic acid, 3-hydroxy- (9CI) (CA INDEX NAME)

OH H2C___CH__(CH2)7__CH__CH2__CO2H

CC 16-4 (Fermentation and Bioindustrial Chemistry)

Section cross-reference(s): 3, 10

ST Pseudomonas recombinant produpolyhydroxyalkanoates carbon sources

IT Genetic engineering

(PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Gene, microbial

RL: BAC (Biological activity or effector, except adverse); BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(PhaG; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Polyesters, biological studies

RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic preparation); BSU (Biological study, unclassified); MFM (Metabolic formation); BIOL (Biological study); FORM (Formation, nonpreparative); PREP (Preparation)

(hydroxycarboxylic acid-based; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Promoter (genetic element)

RL: BAC (Biological activity or effector, except adverse); BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(lacP; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Plasmids

(pBHR86; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Gene, microbial

RL: BAC (Biological activity or effector, except adverse); BPR (Biological process); BSU (Biological study, unclassified); BIOL-(Biological study); PROC (Process)

(phaC1; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Pseudomonas fragi

(recombinant; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT Oxidation

(β-, pathway; PhaG-mediated synthesis of medium chain length poly(3-hydroxyalkanoates) from nonrelated carbon sources in recombinant Pseudomonas fragi)

IT 79-10-7, Acrylic acid, biological studies RL: ADV (Adverse effect, including toxicity); BAC (Biological activity or effector, except adverse); BSU (Biological study,

```
unclassified); BIOL (Biological study)
    (PhaG-mediated synthesis of medium chain length
    poly(3-hydroxyalkanoates) from nonrelated carbon sources in
    recombinant Pseudomonas fragi)
IT 300-85-6DP, 3-Hydroxybutyric acid, copolymer with
  polyhydroxyalkanoates 1883-13-2DP, 3-Hydroxydodecanoic
  acid, copolymer with polyhydroxyalkanoates
  1961-72-4DP, 3-Hydroxytetradecanoic acid, eopolymer with
  polyhydroxyalkanoates 10191-24-9DP, 3-Hydroxyhexanoic
  acid, copolymer with polyhydroxyalkanoates
  14292-26-3DP, 3-Hydroxydecanoic acid, copolymer with
  polyhydroxyalkanoates 14292-27-4DP, 3-Hydroxyoctanoic
  acid, copolymer with polyhydroxyalkanoates
  278185-67-4DP, copolymer with
  polyhydroxyalkanoates
  RL: BMF (Bioindustrial manufacture); BPN (Biosynthetic
  preparation); BSU (Biological study, unclassified); MFM (Metabolic
  formation); BIOL (Biological study); FORM (Formation,
  nonpreparative); PREP (Preparation)
    (PhaG-mediated synthesis of medium chain length
    poly(3-hydroxyalkanoates) from nonrelated carbon sources in
    recombinant Pseudomonas fragi)
  50-99-7, D-Glucose, biological studies 56-81-5,
  1,2,3-Propanetriol, biological studies 77-92-9, biological
  studies 112-80-1, Oleic acid, biological studies 334-48-5,
  Decanoic acid 526-95-4, D-Gluconic acid 12125-02-9, Ammonium
  chloride, biological studies
  RL: BPR (Biological process); BSU (Biological study,
  unclassified); BIOL (Biological study); PROC (Process)
    (PhaG-mediated synthesis of medium chain length
    poly(3-hydroxyalkanoates) from nonrelated carbon sources in
    recombinant Pseudomonas fragi)
IT 215314-08-2
  RL: BAC (Biological activity or effector, except adverse); BPR
  (Biological process); BSU (Biological study, unclassified); BIOL
  (Biological study); PROC (Process)
    (PhaG; PhaG-mediated synthesis of medium chain length
    poly(3-hydroxyalkanoates) from nonrelated carbon sources in
    recombinant Pseudomonas fragi)
IT 134688-88-3
  RL: BAC (Biological activity or effector, except adverse); BPR
  (Biological process); BSU (Biological study, unclassified); BIOL
  (Biological study); PROC (Process)
    (phaC1; PhaG-mediated synthesis of medium chain length
    poly(3-hydroxyalkanoates) from nonrelated carbon sources in
    recombinant Pseudomonas fragi)
REFERENCE COUNT:
                          27 THERE ARE 27 CITED REFERENCES AVAILABLE
                 FOR THIS RECORD. ALL CITATIONS AVAILABLE
                 IN THE RE FORMAT
L64 ANSWER 6 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER:
                           1999:112134 HCAPLUS Full-text
DOCUMENT NUMBER:
                            130:267855
TITLE:
                 Chemical modification of poly(
             hydroxyalkanoates). Copolymers
              bearing pendant sugars
AUTHOR(S):
                     Constantin, Marieta; Simionescu, Cristofer I.;
              Carpov, Adrian; Samain, Eric; Driguez, Hugues
CORPORATE SOURCE:
                           Centre Recherches Macromolecules Vegetales,
              Joseph Fourier Univ., Grenoble, F-38041, Fr.
SOURCE:
                   Macromolecular Rapid Communications (
              1999), 20(2), 91-94
              CODEN: MRCOE3; ISSN: 1022-1336
PUBLISHER:
                     Wiley-VCH Verlag GmbH
DOCUMENT TYPE:
                         Journal
                      English
LANGUAGE:
ED Entered STN: 18 Feb 1999
```

Copolymers of poly(3-hydroxyoctanoates) (PHAs) containing repeating units with unsatd, or brominated pendant side chains were obtained from cultures of Pseudomonas oleovorans grown on mixts, of octanoic acid and undecenoic acid or 11-bromoundecanoic acid as carbon sources. These polymers, bearing reactive functionalities, were used to graft acetylated maltosyl units either by anti-Markovnikov addition to the double bond or SN2 substitution of the halogen. De-O-acetylation of the sugar moieties yielded PHAs with new properties.

IT 201933-09-7P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation);

PROC (Process)

(preparation and characterization of poly(3-hydroxyoctanoates) grafted with acetylated maltose derivs.)

RN 201933-09-7 HCAPLUS

CN 10-Undecenoic acid, 3-hydroxy-, (3R)-, polymer with (3R)-3-hydroxyoctanoic acid, isotactic (9CI) (CA INDEX NAME)

CM I

CRN 198274-26-9 CMF C11 H20 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 $(CH_2)_6$
 CH_2

CM₂

CRN 44987-72-6 CMF C8 H16 O3

Absolute stereochemistry.

- IT 201933-09-7DP, reaction products with acetyl-1-thio-β-

maltose

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(preparation and characterization of poly(3-hydroxyoctanoates) grafted with acetylated maltose derivs.)

RN 201933-09-7 HCAPLUS

CN 10-Undecenoic acid, 3-hydroxy-, (3R)-, polymer with (3R)-3-hydroxyoctanoic acid, isotactic (9Cl) (CA INDEX NAME)

CM I

CRN 198274-26-9 CMF C11 H20 O3

Absolute stereochemistry.

CM 2 .

CRN 44987-72-6 CMF C8 H16 O3

CC 35-8 (Chemistry of Synthetic High Polymers)

ST polyhydroxyalkanoate grafting acetylated maltose

characterization

IT 201933-09-7P 222043-80-3P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(preparation and characterization of poly(3-hydroxyoctanoates) grafted with acetylated maltose derivs.)

IT 2592-37-2DP, reaction products with hydroxyoctanoic

acid-hydroxy-undecenoic acid copolymer 53270-66-9DP,

reaction products with hydroxyoctanoic acid-bromo-

hydroxyundecanoic acid copolymer 201933-09-7DP , reaction products with acetyl-1-thio-β-maltose

222043-80-3DP, reaction products with acetyl-1-S-acetyl-thio-

RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(preparation and characterization of poly(3-hydroxyoctanoates)

grafted with acetylated maltose derivs.)

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L64 ANSWER 7 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

1996:84591 HCAPLUS Full-text

DOCUMENT NUMBER:

124:115514

TITLE:

Production of poly(3-hydroxyalkanoates)

containing aromatic substituents by

Pseudomonas oleovorans

AUTHOR(S):

Curley, Joanne M.; Hazer, Baki; Lenz, Robert

CORPORATE SOURCE:

Department of Polymer Science and Engineering,

University of Massachusetts, Amherst, MA,

01003, USA

SOURCE:

Macromolecules (1996), 29(5), 1762-6

CODEN: MAMOBX; ISSN: 0024-9297

PUBLISHER:

American Chemical Society

DOCUMENT TYPE:

Journal

LANGUAGE:

English

ED Entered STN: 09 Feb 1996

P. oleovorans was grown sep. on 5-(4'-tolyl)valeric acid, 5-(4'-biphenyl)valeric acid, 5-(4'-biphenyl)valeric acid, and 8-(4'-tolyl)octanoic acid AB either as the sole C source or as a cofeed with either nonanoic acid or 5-phenylvaleric acid. For polymer production, 5-(4'-tolyl)valeric acid was the most effective growth substrate of the 5. It resulted in the production of poly-3-hydroxy-5-(4'-tolyl)valerate, a crystalline polymer with a glass transition temperature of 18° and a melting transition of 95°. This poly(3-hydroxyalkanoate) (PHA) is apparently the 1st example of a crystalline aromatic-containing bacterial PHA. When P. oleovorans was cofed an equimolar mixture of 5-phenylvaleric acid and 5-(4'-tolyl)valeric acid, the polymer produced contained 36 mol% of 3-hydroxy-5-phenylvalerate and 64 mol% of 3-hydroxy-5-(4'-tolyl)valerate, and it did not crystallize.

IT 172923-04-5P 172923-06-7P 172923-07-8P

172923-08-9P

RL: BPN (Biosynthetic preparation); PRP (Properties); BIOL

(Biological study); PREP (Preparation)

(production of poly(3-hydroxyalkanoates) containing aromatic substituents

by Pseudomonas oleovorans)

RN 172923-04-5 HCAPLUS

CN Benzenepentanoic acid, β-hydroxy-, (βR)-, homopolymer, isotactic (9CI) (CA INDEX NAME)

CM 1

CRN 153744-07-1

CMF C11 H14 O3

RN 172923-06-7 HCAPLUS

CN Benzenepentanoic acid, β -hydroxy-4-methyl-, (R)-, homopolymer (9Cl) (CA INDEX NAME)

CM I

CRN 172923-05-6 CMF C12 H16 O3

Absolute stereochemistry.

RN 172923-07-8 HCAPLUS

CN Benzenepentanoic acid, β-hydroxy-4-methyl-, (R)-, polymer with (R)-3-hydroxynonanoic acid (9CI) (CA INDEX NAME)

CM I

CRN 172923-05-6 CMF C12 H16 O3

Absolute stereochemistry.

CM 2

CRN 33796-87-1 CMF C9 H18 O3

Absolute stereochemistry.

RN 172923-08-9 HCAPLUS

CN Benzenepentanoic acid, β-hydroxy-4-methyl-, (R)-, polymer with (R)-β-hydroxybenzenepentanoic acid (9CI) (CA INDEX NAME)

CM I

CRN 172923-05-6 CMF C12 H16 O3

Absolute stereochemistry.

CM₂

CRN 153744-07-1 CMF C11 H14 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 Ph

CC 16-4 (Fermentation and Bioindustrial Chemistry)

Section cross-reference(s): 35

ST arom polyhydroxyalkanoate prodn Pseudomonas;

tolyl valerate phenylvalerate hydroxy copolymer

Pseudomonas

IT Pseudomonas oleovorans

(production of poly(3-hydroxyalkanoates) containing aromatic substituents

by Pseudomonas olcovorans)

IT Polyesters, preparation

RL: BPN (Biosynthetic preparation); BIOL (Biological study); PREP

(Preparation)

(hydroxycarboxylic acid-based, production of poly(3-

hydroxyalkanoates) containing aromatic substituents by

Pseudomonas oleovorans)

IT 172923-04-5P 172923-06-7P 172923-07-8P

172923-08-9P

RL: BPN (Biosynthetic preparation); PRP (Properties); BIOL

(Biological study); PREP (Preparation)

(production of poly(3-hydroxyalkanoates) containing aromatic substituents

by Pseudomonas olcovorans)

1T 777-93-5P 51994-31-1P, [1,1'-Biphenyl]-4-pentanoic acid

89326-69-2P, Benzenepentanoic acid, 4-ethyl- 101100-52-1P

RL: BPR (Biological process); BSU (Biological study,

unclassified); SPN (Synthetic preparation); BIOL (Biological

study); PREP (Preparation); PROC (Process)

(production of poly(3-hydroxyalkanoates) containing aromatic substituents

by Pseudomonas oleovorans)

L64 ANSWER 8 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: · DOCUMENT NUMBER:

1997:120979 HCAPLUS Full-text

TITLE:

Scanning electron microscopy of

polyhydroxyalkanoate degradation by

bacteria

AUTHOR(S): Molitoris, H. P.; Moss, S. T.; De Koning, G.

J. M.; Jendrossek, D.

CORPORATE SOURCE: Inst. Botanik, Univ. Regensburg, Regensburg,

D-93040, Germany

SOURCE:

Applied Microbiology and Biotechnology (1996), 46(5/6), 570-579

CODEN: AMBIDG; ISSN: 0175-7598

PUBLISHER: Springer DOCUMENT TYPE: Journal LANGUAGE:

English

ED Entered STN: 21 Feb 1997

Bacterial degradation of sheets of selected polyhydroxyalkanoates by Comamonas sp., Pseudomonas lemoignei and Pseudomonas fluorescens GK13 is reported. Five natural polyhydroxyalkanoates were used, namely poly(3-hydroxybutyrate), poly(3-hydroxyvalerate), a copolymer of 3hydroxybutyrate and 3-hydroxyvalerate, a copolymer of mainly 3-hydroxyoctanoate and minor amts. of 3-hydroxybexanoate, and two rubber-like copolymers of saturated and unsatd. hydroxyalkanoic acids that were modified by electron-beam-induced crosslinking. Each of these polymers was degraded by at least one bacterial strain, the rate of hydrolysis being dependent on the surface area of the polymer exposed to attack. SEM of partially degraded samples showed that hydrolysis started at the surface and at phys. lesions in the polymer and proceeded to the inner part of the material. No evidence for areas of non-degradable polymer was found for any of the polymers analyzed, even if the polymer contained chemical

IT 128971-78-8

RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process) (SEM of polyhydroxyalkanoate degradation by bacteria)

RN 128971-78-8 HCAPLUS

CN 7-Octenoic acid, 3-hydroxy-, (R)-, polymer with

(R)-3-hydroxyhexanoic acid, (R)-3-hydroxy-5-hexenoic acid and

(R)-3-hydroxyoctanoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 119003-50-8 CMF C8 H14 O3

Absolute stereochemistry.

CM₂

CRN 119003-49-5 CMF C6 H10 O3

Absolute stereochemistry.

CM₃

CRN 77877-35-1 CMF C6 H12 O3

Absolute stereochemistry. Rotation (-).

CM 4

CRN 44987-72-6 CMF C8 H16 O3

CC 10-2 (Microbial, Algal, and Fungal Biochemistry)

Section cross-reference(s): 35

polyhydroxyalkanoate degrdn bacteria SEM

Bacteria (Eubacteria)

Comamonas

Pseudomonas

Scanning electron microscopy

(SEM of polyhydroxyalkanoate degradation by bacteria)

IT Decomposition

(biodegrdn.; SEM of polyhydroxyalkanoate degradation by bacteria)

IT Polymer degradation

(biol.; SEM of polyhydroxyalkanoate degradation by

IT Polyesters, biological studies

RL: BPR (Biological process); BSU (Biological study,

unclassified); RCT (Reactant); BIOL (Biological study); PROC

(Process); RACT (Reactant or reagent)

(hydroxycarboxylic acid-based; SEM of

polyhydroxyalkanoate degradation by bacteria)

IT 128971-78-8 141455-97-2 141901-08-8 154994-48-6

155075-32-4 160555-53-3 262373-27-3

RL: BPR (Biological process); BSU (Biological study,

unclassified); BIOL (Biological study); PROC (Process)

(SEM of polyhydroxyalkanoate degradation by bacteria):

L64 ANSWER 9 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

1996:352644 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER:

125:53235

TITLE:

Sequential production of two different

polyesters in the inclusion bodies of

Pseudomonas oleovorans

AUTHOR(S):

Curley, Joanne M.; Lenz, Robert W.; Fuller, R.

Clinton

CORPORATE SOURCE:

Dep. Polym. Sci. Eng., Univ. Massachusetts,

Amherest, MA, 01003, USA

SOURCE:

International Journal of Biological

Macromolecules (1996), 19(1), 29-34

CODEN: IJBMDR; ISSN: 0141-8130

PUBLISHER: DOCUMENT TYPE:

Elsevier Journal

LANGUAGE:

English

ED Entered STN: 18 Jun 1996

When Pseudomonas oleovorans was grown on a mixture of 5-phenylvaleric acid, PVA, and nonanoic acid, NA, the reserve polyester produced included both a homopolymer and a copolymer. The homopolymer poly-3-hydroxy-5- phenylvalerate, PHPV, contained only 3-hydroxy-5phenylvalerate units, while the copolymer contained the same long-chain 3-hydroxyalkanoates as those present in the copolymer poly-3hydroxynonanoate, PHN, which is produced from nonanoic acid alone. The intracellular location of each of these polymers was determined by selective staining of the inclusion body granules with ruthenium tetraoxide and examination by transmission electron microscopy showed that both types of polyesters occurred in the same granule. PHN was present in the center of the granule, while PHPV accumulated around the PHN in the inclusion body. The proteins associated with the inclusion bodies wee separated using sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE). In all cases, two different polymerase enzymes of mol. weight 59 and 55 KDa were present, indicating that the same polymerase enzyme system was responsible for the production of both PHN and PHPV. Attempts were made to produce a random copolymer containing both alkyl and phenylalkyl repeat units by varying the growth conditions, but a mixture of PHN and PHPV was always produced instead.

IT 134736-36-0P

RL: BOC (Biological occurrence); BSU (Biological study,

unclassified); MFM (Metabolic formation); PRP (Properties); PUR

(Purification or recovery); BIOL (Biological study); FORM

(Formation, nonpreparative); OCCU (Occurrence); PREP (Preparation)

(sequential production of 2 different polyesters in the inclusion

bodies of Pseudomonas oleovorans)

RN 134736-36-0 HCAPLUS

CN Benzenepentanoic acid, β-hydroxy-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 41479-99-6 CMF C11 H14 O3

```
Ph_CH2_CH2_CH_CH2_CO2H
```

CC 10-2 (Microbial, Algal, and Fungal Biochemistry) ST polyester sequential formation Pseudomonas inclusion body; polyhydroxyalkanoate sequential formation Pseudomonas inclusion body

IT Inclusion bodies

Pseudomonas oleovorans

(sequential production of 2 different polyesters in the inclusion bodies of Pseudomonas olcovorans)

IT 134688-88-3, Hydroxyalkanoate polymerase

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); BIOL (Biological study) (sequential production of 2 different polyesters in the inclusion bodies of Pseudomonas oleovorans)

IT 120659-39-4P 134736-36-0P

RL: BOC (Biological occurrence); BSU (Biological study, unclassified); MFM (Metabolic formation); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); FORM (Formation, nonpreparative); OCCU (Occurrence); PREP (Preparation) (sequential production of 2 different polyesters in the inclusion bodies of Pseudomonas oleovorans)

IT 112-05-0, Nonanoic acid 2270-20-4, 5-Phenylvaleric acid RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process) (sequential production of 2 different polyesters in the inclusion bodies of Pseudomona's oleovorans)

L64 ANSWER 10 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER:

1990:516180 HCAPLUS Full-text

DOCUMENT NUMBER:

113:116180

TITLE:

Physical characteristics of poly(3-hydroxyalkanoates) and poly(3-hydroxyalkenoates) produced by Pseudomonas oleovorans grown on

aliphatic hydrocarbons

AUTHOR(S):

Preusting, Hans; Nijenhuis, Atze; Witholt,

Bernard

CORPORATE SOURCE: Groningen Biotechnol. Cent., Univ. Groningen,

Groningen, 9747 AG, Neth.

SOURCE:

Macromolecules (1990), 23(19),

4220-4

CODEN: MAMOBX; ISSN: 0024-9297

DOCUMENT TYPE:

Journal LANGUAGE: ED Entered STN: 29 Sep 1990

Pseudomonas oleovorans accumulated poly(3- hydroxyalkanoates) (PHAs) after growth on n-alkanes and 1-alkenes. The composition and phys. characteristics of these polyesters were substrate dependent. When n-alkanes (n-hexane to n-decane) were used, PHAs were formed consisting of 3-hydroxyalkanoate monomers of which the pendant group varied from a Me to a heptyl group (saturated PHAs). When 1-alkenes (1-octene and 1decene) were used as the carbon source, the polyester consisted of both 3-hydroxyalkanoate and terminally unsatd. 3-hydroxyalkanoate monomers, of which the pendant group varied in length between a Pr and a heptyl group (unsatd. PHAs). The structure of the PHAs was confirmed by 1H and 13C NMR. Apart from the PHAs isolated from n-hexane and 1-alkene grown cells, the copolymers were partly crystalline ($\Delta hm = 6.6-18.7 \text{ J/g}$), having m.ps. which varied between 38.9 and 58.5° and glass temps. (Tg) from -30.8 to -39.7°. The Tg's of the amorphous polymers ranged from -25.8 to -43.1°. The mol. wts. of the isolated polymers ranged from 178,000 to 330,000.

IT 128971-78-8P 128999-53-1P

RL: BPN (Biosynthetic preparation); PRP (Properties); BIOL

(Biological study); PREP (Preparation)

(biochem, preparation and structure of, from Pseudomonas oleovorans, effect of feed substrate on)

RN 128971-78-8 HCAPLUS

CN 7-Octenoic acid, 3-hydroxy-; (R)-, polymer with

(R)-3-hydroxyhexanoic acid, (R)-3-hydroxy-5-hexenoic acid and (R)-3-hydroxyoctanoic acid (9Cl) (CA INDEX NAME) .

CM I

CRN 119003-50-8 CMF C8 H14 O3

Absolute stereochemistry.

CM 2

CRN 119003-49-5 CMF C6 H10 O3

Absolute stereochemistry.

CM 3

CRN 77877-35-1 CMF C6 H12 O3

Absolute stereochemistry. Rotation (-).

CM 4

CRN 44987-72-6 CMF C8 H16 O3

Absolute stereochemistry.

RN 128999-53-1 HCAPLUS

CN 9-Decenoic acid, 3-hydroxy-, (R)-, polymer with

(R)-3-hydroxydecanoic acid, (R)-3-hydroxyhexanoic acid,

(R)-3-hydroxy-5-hexenoic acid, (R)-3-hydroxynonanoic acid,

(R)-3-hydroxyoctanoic acid and (R)-3-hydroxy-7-octenoic acid (9Cl)

(ĆA INDEX NAME)

CM 1

CRN 119003-52-0 CMF C10 H18 O3 Absolute stereochemistry.

CM 2

CRN 119003-50-8 CMF C8 H14 O3

Absolute stereochemistry.

$$HO_2C$$
 R
 $(CH_2)_3$
 CH_2

CM 3

CRN 119003-49-5 CMF C6 H10 O3

Absolute stereochemistry.

CM 4

CRN 77877-35-1 CMF C6 H12 O3

Absolute stereochemistry. Rotation (-).

CM 5

CRN 44987-72-6 CMF C8 H16 O3

CM 6

CRN 33796-87-1 CMF C9 H18 O3

Absolute stereochemistry.

CM 7

CRN 19525-80-5 CMF C10 H20 O3

Absolute stereochemistry. Rotation (-).

CC 36-2 (Physical Properties of Synthetic High Polymers) Section cross-reference(s): 35

ST Pseudomonas olcovorans polyhydroxyalkanoate structure property

IT Polyesters, preparation

RL: BPN (Biosynthetic preparation); BIOL (Biological study); PREP (Preparation)

(biochem. preparation of, from Pseudomonas oleovorans, feed substrate effect on structure in relation to)

IT Crystallinity

Glass temperature and transition (of biochem, prepared poly(hydroxyalkanoates

IT Pseudomonas oleovorans

(polyhydroxyalkanoates produced from, structure of, feed substrate effect on)

IT 128971-75-5P 128971-76-6P 128971-77-7P 128971-78-8P 128999-52-0P 128999-53-1P 174793-36-3P

RL: BPN (Biosynthetic preparation); PRP (Properties); BIOL

(Biological study); PREP (Preparation)

(biochem. preparation and structure of, from Pseudomonas oleovorans, effect of feed substrate on)

IT 110-54-3, Hexane, properties 111-65-9, Octane, properties 111-66-0, 1-Octene 111-84-2, Nonane 124-18-5, Decane

142-82-5, Heptane, properties 872-05-9, 1-Decene

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); PRP (Properties); BIOL

(Biological study)

(substrate, for biochem. production of poly(hydroxyalkanoates) from Pseudomonas oleovorans, structure in relation to)

L64 ANSWER 11 OF 15 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 1989:91797 FICAPLUS Full-text

DOCUMENT NUMBER:

110:91797

TITLE:

Formation of polyesters by Pseudomonas oleovorans: effect of substrates on formation and composition of poly-(R)-3-

hydroxyalkanoates and poly-(R)-3-

hydroxyalkenoates

AUTHOR(S):

Lageveen, Roland G.; Huisman, Gjalt'W.;

Preusting, Hans; Ketelaar, Peter; Eggink,

Gerrit; Witholt, Bernard

CORPORATE SOURCE: Groningen Biotechnol. Cent., Univ. Groningen,

Groningen, 9747 AG, Neth.

SOURCE:

Applied and Environmental Microbiology (

1988), 54(12), 2924-32

CODEN: AEMIDF; ISSN: 0099-2240

DOCUMENT TYPE: LANGUAGE:

Journal

ED Entered STN: 17 Mar 1989

P. oleovorans grows on C6 to C12 n-alkanes and 1-alkenes. These substrates are oxidized to the corresponding fatty acids, which are oxidized further via the β-oxidation pathway, yielding shorter fatty acids which have lost one or more C2 units. P. oleovorans normally utilizes β-oxidation pathway intermediates for growth, but in this study, the intermediate 3-hydroxy fatty acids can also be polymerized to intracellular poly-(R)-3hydroxyalkanoates (PHAs) when the medium contains limiting amts, of essential elements, such as nitrogen. The monomer composition of these polyesters is a reflection of the substrates used for growth of P. oleovorans. The largest monomer found in PHAs always contained as many C atoms as did the n-alkane used as a substrate. Monomers which were shorter by one or more C2 units were also observed Thus, for C-even substrates, only C-even monomers were found, with (R)-3-hydroxyheptanoate as the smallest monomer. 1-Alkenes were also incorporated into PHAs, albeit less efficiently and with lower yields than n-alkanes. These PHAs contained both saturated and unsatd, monomers, apparently because the 1-alkene substrates could be oxidized to carboxylic acids at either the saturated or the unsatd, ends. Up to 55% of the PHA monomers contained terminal double bonds when P. oleovorans was grown on 1-alkenes. The degree of unsatn, of PHAs could be modulated by varying the ratio of alkenes to alkanes in the growth medium. Since I-alkenes were also shortened before being polymerized, as was the case for n-alkanes, copolymers which varied with respect to both monomer chain length and the percentage of terminal double bonds were formed during nitrogenlimited growth of P. oleovorans on 1-alkenes. Such polymers are expected to be useful for future chemical modifications.

IT 119003-49-5 119003-50-8 119003-51-9

119003-52-0

RL: BIOL (Biological study)

(in polyester formed by Pseudomonas oleovorans)

RN 119003-49-5 HCAPLUS

CN 5-Hexenoic acid, 3-hydroxy-, (3R)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 119003-50-8 HCAPLUS

CN 7-Octenoic acid, 3-hydroxy-, (R)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 119003-51-9 HCAPLUS

CN 8-Nonenoic acid, 3-hydroxy-, (3R)- (CA INDEX NAME)

Absolute stereochemistry.

RN 119003-52-0 HCAPLUS

CN 9-Decenoic acid, 3-hydroxy-, (R)- (9CI) (CA INDEX NAME)

CC 10-2 (Microbial Biochemistry)

ST polyester formation Pseudomonas

polyhydroxyalkanoate polyhydroxyalkenoate

IT Polyesters, biological studies

RL: BIOL (Biological study)

(in polyester formed by Pseudomonas oleovorans)

IT Alkanes, biological studies

Alkenes, biological studies

RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process) (metabolism of, by Pseudomonas oleovorans, polyesters

formation in relation to)

IT Pseudomonas oleovorans

 $(polyhydroxyalkanoates\ and\ polyhydroxyalkenoates$

formation by, substrates effect on)

IT 7787-35-1 19525-80-5 28254-78-6 33796-87-1 44987-72-6

85233-44-9 97961-62-1 119003-49-5 119003-50-8

119003-51-9 119003-52-0

RL: BIOL (Biological study)

(in polyester formed by Pseudomonas oleovorans)

IT 110-54-3, Hexane, biological studies 111-65-9, Octane,

biological studies 111-66-0, 1-Octene 111-84-2, Nonane

112-40-3, Dodecane 112-41-4, 1-Dodecene 124-11-8, 1-Nonene

124-18-5, Decane 142-82-5, Heptane, biological studies

592-41-6, 1-Hexene, biological studies 872-05-9, 1-Decene

1120-21-4, Undecane

RL: BPR (Biological process); BSU (Biological study,

unclassified); BIOL (Biological study); PROC (Process)

(metabolism of, by Pseudomonas oleovorans, polyesters

formation in relation to)

=> d 164 12-15 ibib ed ab hit ind

L64 ANSWER 12 OF 15 MEDLINE on STN

ACCESSION NUMBER: 91175643 MEDLINE Full-text

DOCUMENT NUMBER: PubMed ID: 2078535

TITLE: Production of unsaturated polyesters by Pseudomonas oleovorans.

AUTHOR: Fritzsche K; Lenz R W; Fuller R C

CORPORATE SOURCE: Department of Polymer Science and Engineering,

University of Massachusetts, Amherst 01003.

SOURCE: International journal of biological macromolecules,

(1990 Apr) Vol. 12, No. 2, pp. 85-91.

Journal code: 7909578. ISSN: 0141-8130.

PUB. COUNTRY: ENGLAND: United Kingdom

DOCUMENT TYPE: Journal; Article; (JOURNAL ARTICLE)

(RESEARCH SUPPORT, U.S. GOVT, NON-P.H.S.)

LANGUAGÈ:

English

FILE SEGMENT: Priority Journals

ENTRY MONTH: 199105

ENTRY DATE: Entered STN: 19 May 1991

Last Updated on STN: 19 May 1991

Entered Medline: 1 May 1991

ED Entered STN: 19 May 1991

Last Updated on STN: 19 May 1991

Entered Medline: 1 May 1991

AB Pseudomonas oleovorans was grown separately on 3-hydroxy-6-octenoic acid and 3-hydroxy-7-octenoic acid as the only carbon source and under ammonium nutrient-limiting conditions to produce storage polyesters. The polyesters produced contained mainly unsaturated C8 units. Small amounts of both the saturated and the unsaturated C6 units were also present, but only about 1% of the saturated 3-hydroxyoctanoate units was detected. The polyester obtained from 3-hydroxy-6-octenoic acid, which was a mixture of the cis and trans isomers, also contained units with cis and trans double bonds. The weight average molecular weights of the polymers produced were in the range of 339,000-383,000 as determined by g.p.c. relative to polystyrene, with Mw/Mn ratios of 1.8-2.1. The mechanism of PHA formation from n-octene previously reported is discussed in relation to the present results, and the two were found to be in good agreement.

T1 Production of unsaturated polyesters by Pseudomonas

oleovorans.

Pseudomonas oleovorans was grown separately on 3-hydroxy-6-octenoic acid and 3-hydroxy-7-octenoic acid as the only carbon source and under ammonium nutrient-limiting conditions to produce storage polyesters. The polyesters produced contained mainly unsaturated C8 units. Small amounts of both the saturated and the unsaturated C6 units were also present, but only about 1% of the saturated 3-hydroxyoctanoate units was detected. The polyester obtained from 3-hydroxy-6-octenoic acid, which was a mixture of the cis and trans isomers, also contained units with cis and trans double bonds. The weight average molecular weights of the polymers produced were in the range of 339,000-383,000 as determined by g.p.c. relative to polystyrene, with Mw/Mn ratios of 1.8-2.1. The mechanism of PHA formation from n-octene previously reported is discussed in relation to the present results, and the two were found to be in good agreement.

CT Culture Media

*Fatty Acids, Monounsaturated: ME, metabolism

*Hydroxy Acids: ME, metabolism

Macromolecular Substances

Magnetic Resonance Spectroscopy

Molecular Weight

*Polymers: ME, metabolism

Pseudomonas: GD, growth & development

*Pseudomonas: ME, metabolism

RN 120676-01-9 (3-hydroxy-7-octenoic acid); 128940-64-7

(3-hydroxy-6-octenoic acid)

CT Culture Media

*Fatty Acids, Monounsaturated: ME, metabolism

*Hydroxy Acids: ME, metabolism

Macromolecular Substances

Magnetic Resonance Spectroscopy

Molecular Weight

*Polymers: ME, metabolism

Pseudomonas: GD, growth & development

*Pseudomonas: ME, metabolism

RN 120676-01-9 (3-hydroxy-7-octenoic acid); 128940-64-7

(3-hydroxy-6-octenoic acid)

CN 0 (Culture Media); 0 (Fatty Acids, Monounsaturated); 0 (Hydroxy

Acids); 0 (Macromolecular Substances); 0 (Polymers)

L64 ANSWER 13 OF 15 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 2005:401806 BIOSIS Full-text

DOCUMENT NUMBER: PREV200510189913

TITLE: Mol

Molecular characterization of extracellular medium-chain-length poly(3-hydroxyalkanoate)

depolymerase genes from Pseudomonas

alcaligenes strains.

AUTHOR(S): Kim, Do Young; Kim, Hyun Chul; Kim, Sun Young;

Rhee, Young Ha [Reprint Author]

CORPORATE SOURCE: Chungnam Natl Univ, Sch Biosci and Biotechnol, Dept

Microbiol, Taejon 305764, South Korea

yhrhee@cnu.ac.kr

SOURCE: Journal of Microbiology, (JUN 2005) Vol. 43, No. 3,

pp. 285-294. ISSN: 1225-8873.

DOCUMENT TYPE: Article

LANGUAGE:

English

ENTRY DATE: Entered STN: 5 Oct 2005

Last Updated on STN: 5 Oct 2005

ED Entered STN: 5 Oct 2005

Last Updated on STN: 5 Oct 2005

Ab acterial strain M4-7 capable of degrading various polyesters, such as poly(c-caprolactone), poly(3-hydroxybutyrate-co-3- hydroxyvalerate), poly(3-hydroxyoctanoate), and poly(3-hydroxy-5-phenylvalerate), was isolated from a marine environment and identified as Pseudomonas alcaligenes. The relative molecular mass of a purified extracellular medium-chain-length poly(3-hydroxyalkanoate) (MCL-PRA) depolymerase (PhaZ(PalM4-7)) from P. alcaligenes M4-7 was 28.0 kDa, as determined by SDS-PAGE. The PhaZ(PalM4-7) was most active in 50 mM glycine-NaOH buffer (pH 9.0) at 35 degrees C. It was insensitive to dithiothreitol, sodium azide, and iodoacetamide, but susceptible to p-hydroxymercuribenzoic acid, N-bromosuccinimide, acetic anhydride, EDTA, diisopropyl fluorophosphate, phenylmethylsulfonyl fluoride, Tween 80, and Triton X-100. In this study, the genes encoding MCL-PHA depolymerase were cloned, sequenced, and characterized from a soil bacterium, P alcaligenes LB19 (Kim et al., 2002, Biomacromolecules 3, 291-296) as well as P alcaligenes M4-7. The structural gene (PhaZ(PalLB19)) of MCL-PHA depolymerase of P alcaligenes LB19 consisted of an 837 bp open reading frame (ORF) encoding a protein of 278 amino acids with a deduced M-r of 30,188 Da. However, the MCL-PHA depolymerase gene (phaZ(PalM4-7)) of P. alcaligenes M4-7 was composed of an 834 bp ORF encoding a protein of 277 amino acids with a deduced M-r of 30,323 Da. Amino acid sequence analyses showed that, in the two different polypeptides, a substrate-binding domain and a catalytic domain are located in the N-terminus and in the C-terminus, respectively. The PhaZ(PalLB19) and the PhaZ(PalM4-7) commonly share the lipase box, GISSG in their catalytic domains, and utilize (111)Asn and (110)Ser residues, respectively, as oxyanions that play an important role in transition-state stabilization of hydrolytic reactions.

TI Molecular characterization of extracellular medium-chain-length

poly(3-hydroxyalkanoate) depolymerase genes from

Pseudomonas alcaligenes strains.

AB A bacterial strain M4-7 capable of degrading various polyesters, such as poly(c-caprolactone), poly(3-hydroxybutyrate-co-3- hydroxyvalerate), poly(3-hydroxyoctanoate), and poly(3-hydroxy-5-phenylvalerate), was isolated from a marine environment and identified as Pseudomonas alcaligenes. The relative molecular mass of a purified extracellular medium-chain-length poly(3-hydroxyalkanoate) (MCL-PRA) depolymerase

10/531,689

(PhaZ(PalM4-7)) from P. alcaligenes M4-7 was 28.0 kDa, as determined by SDS-PAGE. The PhaZ(PalM4-7) was most active in 50 mM glycine-NaOH buffer (pH 9.0) at 35 degrees C. It was insensitive to dithiothreitol, sodium azide, and iodoacetamide, but susceptible to phydroxymercuribenzoic acid, N-bromosuccinimide, acetic anhydride, EDTA, diisopropyl fluorophosphate, phenylmethylsulfonyl fluoride, Tween 80, and Triton X-100. In this study, the genes encoding MCL-PHA depolymerase were cloned, sequenced, and characterized from a soil bacterium, P alcaligenes LB19 (Kim et al., 2002, Biomacromolecules 3, 291-296) as well as P alcaligenes M4-7. The structural gene (PhaZ(PalLB19)) of MCL-PHA depolymerase of P alcaligenes LB19 consisted of an 837 bp open reading frame (ORF) encoding a protein of 278 amino acids with a deduced M-r of 30,188 Da. However, the MCL-PHA depolymerase gene (phaZ(PalM4-7)) of P alcaligenes M4-7 was composed of an 834 bp ORF encoding a protein of 277 amino acids with a deduced M-r of 30,323 Da. Amino acid sequence analyses showed that, in the two different polypeptides, a substrate-binding domain and a catalytic domain are located in the N-terminus and in the C-terminus, respectively. The PhaZ(PalLB19) and the PhaZ(PalM4-7) commonly share the lipase box, GISSG in their catalytic domains, and utilize (111)Asn and (110)Ser residues, respectively, as oxyanions that play an important role in transition-state stabilization of hydrolytic reactions.

```
ORGN Classifier
    Pseudomonadaceae 06508
   Super Taxa
    Gram-Negative Aerobic Rods and Cocci; Eubacteria; Bacteria;
    Microorganisms
   Organism Name
      Pseudomonas alcaligenes (species): strain-LB19,
    strain-M4-7
  Taxa Notes
    Bacteria, Eubacteria, Microorganisms
RN 60-00-4 (EDTA)
   108-24-7 (acetic anhydride)
   3483-12-3 (dithiothreitol)
   9002-93-1 (Triton X-100)
   26628-22-8 (sodium azide)
   128-08-5 (N-bromosuccinimide)
   9005-65-6 (Tween 80)
   55-91-4 (diisopropyl fluorophosphate)
   144-48-9 (iodoacetamide)
   329-98-6 (phenylmethylsulfonyl fluoride)
    134736-36-0 (poly(3-hydroxy-5-phenylvalerate))
   1126-48-3 (p-hydroxymercuribenzoic acid)
GEN Pseudomonas alcaligenes MCL-PHA gene (Pseudomonadaceae)
CC Genetics - General 03502
   Biochemistry studies - General 10060
   Physiology and biochemistry of bacteria 31000
   Genetics of bacteria and viruses 31500
1T Major Concepts
    Molecular Genetics (Biochemistry and Molecular Biophysics)
   Chemicals & Biochemicals
    EDTA; acetic anhydride; dithiothreitol; Triton X-100; sodium
    azide; N-bromosuccinimide; Tween 80; diisopropyl
    fluorophosphate; iodoacetamide; phenylmethylsulfonyl fluoride;
    poly(3-hydroxybutyrate-co-3-hydroxyvalerate);
    poly(3-hydroxyoctanoate); poly(alpha-caprolactone);
    poly(3-hydroxy-5-phenylvalerate); poly(3-hydroxyalkanoates);
    p-hydroxymercuribenzoic acid
IT Methods & Equipment
    SDS-polyacrylamide gel electrophoresis [SDS-PAGE]:
    electrophoretic techniques, laboratory techniques
   Miscellaneous Descriptors
    molecular ccaracterization
ORGN Classifier
    Pseudomonadaceae 06508
   Super Taxa
    Gram-Negative Aerobic Rods and Cocci; Eubacteria; Bacteria;
    Microorganisms
   Organism Name
      Pseudomonas alcaligenes (species): strain-LB19,
    strain-M4-7
   Taxa Notes
    Bacteria, Eubacteria, Microorganisms
RN 60-00-4 (EDTA)
   108-24-7 (acetic anhydride)
   3483-12-3 (dithiothreitol)
   9002-93-1 (Triton X-100)
   26628-22-8 (sodium azide)
   128-08-5 (N-bromosuccinimide)
  .9005-65-6 (Tween 80)
   55-91-4 (diisopropyl fluorophosphate)
   144-48-9 (iodoacetamide)
   329-98-6 (phenylmethylsulfonyl fluoride)
    134736-36-0 (poly(3-hydroxy-5-phenylvalerate))
```

1126-48-3 (p-hydroxymercuribenzoic acid) GEN Pseudomonas alcaligenes MCL-PHA gene (Pseudomonadaceae)

L64 ANSWER 14 OF 15 BIOSIS COPYRIGHT (c) 2007 The Thomson

Corporation on STN

ACCESSION NUMBER: 1999:526726 BIOSIS Full-text

DOCUMENT NUMBER: PREV199900526726

TITLE: Intracellular depolymerase activity in isolated

inclusion bodies containing

polyhydroxyalkanoates with long alkyl and

functional substituents in the side chain.

Foster, L. J. R.; Lenz, R. W.; Fuller, R. C.

[Reprint author]

CORPORATE SOURCE: Department of Biochemistry and Molecular Biology,

University of Massachusetts, Amherst, MA, 01003,

USA

SOURCE:

International Journal of Biological Macromolecules,

(Nov., 1999) Vol. 26, No. 2-3, pp. 187-192. print.

CODEN: IJBMDR. ISSN: 0141-8130.

DOCUMENT TYPE: Article LANGUAGE: English

Entered STN: 10 Dec 1999 **ENTRY DATE:**

Last Updated on STN: 5 Jun 2000

ED Entered STN: 10 Dec 1999 Last Updated on STN: 5 Jun 2000

The in vitro degradation of isolated Pseudomonas oleovorans inclusion bodies containing either poly-3- hydroxynonanoate (PHN), or poly(-3hydroxy-5-phenylvalerate) (PHPV), or a mixture of these two polymers was investigated. When incubated at 30degreeC and pH 9, inclusion bodies containing either polyhydroxyoctanoate (PHO), PHN or PHPV exhibited similar degradation rates of approximately 0.94 (+-3%) mg/h. The PHN and PHPV components for inclusion bodies containing a mixture of PHN and PHPV showed similar degradation rates; that is the ratios showed little change and remained at approximately 50 weight% (+-3%) for each component. These results contrast markedly with in vivo studies for similar inclusion bodies in whole cells. The results suggest that the synthesis and degradation of these novel polyhydroxyalkanoates by P. oleovorans proceeds by the same enzymatic pathway. In addition, comparisons between the in vivo and in vitro polymer degradation suggest that the activity of the intracellular depolymerase does not control the rate limiting step of PHPV degradation in vivo. Instead, the presence of an aromatic group in the repeating units of this polymer may inhibit the utilization of the monomeric units of PHPV as a reserve carbon source by the cells.

TI Intracellular depolymerase activity in isolated inclusion bodies containing polyhydroxyalkanoates with long alkyl and functional substituents in the side chain.

The in vitro degradation of isolated Pseudomonas oleovorans inclusion bodies containing either poly-3- hydroxynonanoate (PHN), or poly(-3hydroxy-5-phenylvalerate) (PHPV), or a mixture of these two polymers was investigated. When incubated at 30degreeC and pH 9, inclusion bodies containing either polyhydroxyoctanoate (PHO), PHN or PHPV exhibited similar degradation rates of approximately 0.94 (+-3%) mg/h. The PHN and PHPV components for inclusion bodies containing a mixture of PHN and PHPV showed similar degradation rates; that is the ratios showed little change and remained at approximately 50 weight% (+-3%) for each component. These results contrast markedly with in vivo studies for similar inclusion bodies in whole cells. The results suggest that the synthesis and degradation of these novel polyhydroxyalkanoates by P. oleovorans proceeds by the same enzymatic pathway. In addition, comparisons between the in vivo and in vitro polymer degradation suggest that the activity of the intracellular depolymerase does not control the rate limiting step of PHPV degradation in vivo. Instead, the presence of an aromatic group in the repeating units of this polymer may inhibit the utilization of the monomeric units of PHPV as a reserve carbon source by the

Major Concepts

Enzymology (Biochemistry and Molecular Biophysics)

Parts, Structures, & Systems of Organisms

inclusion bodies

Chemicals & Biochemicals

depolymerase: intracellular; poly(-3-hydroxy-5-phenylvalerate)

[PHPV]; poly-3-hydroxyonanoate [PHN];

polyhydroxyalkanoates

ORGN Classifier

Pseudomonadaceae 06508

Super Taxa

Gram-Negative Aerobic Rods and Cocci; Eubacteria; Bacteria;

Microorganisms

Organism Name

Taxa Notes

Pseudomonas oleovorans

Bacteria, Eubacteria, Microorganisms

9030-73-3 (depolymerase)

134736-36-0 (poly(-3-hydroxy-5-phenylvalerate))

134736-36-0 (PHPV) CC Enzymes - General and comparative studies: coenzymes 10802

Biochemistry studies - General 10060

Physiology and biochemistry of bacteria 31000

Food microbiology - General and miscellaneous 39008

IT - Major Concepts

Enzymology (Biochemistry and Molecular Biophysics)

```
IT Parts, Structures, & Systems of Organisms
      inclusion bodies
    Chemicals & Biochemicals
      depolymerase: intracellular; poly(-3-hydroxy-5-phenylvalerate)
      [PHPV]; poly-3-hydroxyonanoate [PHN];
      polyhydroxyalkanoates
    Miscellaneous Descriptors
      industrial microbiology
ORGN Classifier
       Pseudomonadaceae 06508
    Super Taxa
       Gram-Negative Aerobic Rods and Cocci; Eubacteria; Bacteria;
       Microorganisms
    Organism Name
        Pseudomonas oleovorans
    Taxa Notes
       Bacteria, Eubacteria, Microorganisms
RN 9030-73-3 (depolymerase)
      134736-36-0 (poly(-3-hydroxy-5-phenylvalerate))
      134736-36-0 (PHPV)
1.64 ANSWER 15 OF 15 BIOSIS COPYRIGHT (c) 2007 The Thomson
    Corporation on STN
ACCESSION NUMBER: 1999:39335 BIOSIS Full-text
DOCUMENT NUMBER: PREV199900039335
TITLE:
                       Isolation of an aromatic
                 polyhydroxyalkanoates-degrading bacterium.
AUTHOR(S):
                            Ju, He-Sug; Kim, Jungho; Kim, Hoon [Reprint author]
CORPORATE SOURCE: Dep. Agric. Chem., Sunchon Natl. Univ., Sunchon
                 540-742, South Korea
SOURCE:
                         Journal of Microbiology and Biotechnology, (Oct.,
                 1998) Vol. 8, No. 5, pp. 540-542. print.
                 ISSN: 1017-7825.
DOCUMENT TYPE:
                                    Article
LANGUAGE:
                              English
ENTRY DATE:
                               Entered STN: 3 Feb 1999
                 Last Updated on STN: 3 Feb 1999
ED Entered STN: 3 Feb 1999
    Last Updated on STN: 3 Feb 1999
             Five microorganisms capable of degrading an aromatic medium-chain-length polyhydroxyalkanoate (PHAMCL), poly(3-hydroxy-5-
            phenylvalerate) (PHPV), were isolated from wastewater-treatment sludge. Among the isolates, ISO2 showed degrading activity consistently during
             several transfers. The isolate JS02 could hydrolyze another aromatic MCL copolyester, poly(3-hydroxy-5phenoxyvalerate-co-3-hydroxy-7-
            phenoxyheptanoate), (P (5POHV-co-7POHH)), and other short-chain-length PHAs (PHASCL) such as poly(3-hydroxybutyrate) (P3(HB)), poly(3-hydroxybutyrate)
             hydroxybutyrate- co-4-hydroxybutyrate) (P(3HB-co-4HB)), and poly(3-hydroxybutyrate- co-3-hydroxyvalerate) (P(3HB-co-3HV)) with relatively
             low activity. The culture supernatant of JS02 showed hydrolyzing activity for the p-nitrophenyl esters of fatty acids.
Tl Isolation of an aromatic polyhydroxyalkanoates-degrading
    bacterium.
            Five microorganisms capable of degrading an aromatic medium-chain-length polyhydroxyalkanoate (PHAMCL), poly(3-hydroxy-5-
             phenylvalerate) (PHPV), were isolated from wastewater-treatment sludge. Among the isolates, JS02 showed degrading activity consistently during
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             hydroxybutyrate- co-4-hydroxybutyrate) (P(3HB-co-4HB)), and poly(3-hydroxybutyrate- co-3-hydroxyvaterate) (P(3HB-co-3HV)) with relatively
             low activity. The culture supernatant of JS02 showed hydrolyzing activity for the p-nitrophenyl esters of fatty acids.
IT Major Concepts
       Bacteriology; Metabolism; Methods and Techniques
     Chemicals & Biochemicals
       aromatic polyhydroxyalkanoate: degradation,
       isolation; poly(3-hydroxy-5-phenylvalerate)
RN 134736-36-0 (poly(3-hydroxy-5-phenylvalerate))
CC Microbiological apparatus, methods and media 32000
    Biochemistry methods - General 10050
    Biophysics - Methods and techniques 10504
    Metabolism - General metabolism and metabolic pathways 13002
     Physiology and biochemistry of bacteria 31000
IT Major Concepts
       Bacteriology; Metabolism; Methods and Techniques
IT Chemicals & Biochemicals
       aromatic polyhydroxyalkanoate: degradation,
       isolation; poly(3-hydroxy-5-phenylvalerate)
IT Methods & Equipment
       bacteria isolation: Isolation/Purification Techniques: CT,
```

isolation method; bacterial culture; cell culture techniques, culture method; gas chromatography; chromatographic techniques,

isolation method

ORGN Classifier
Bacteria 05000
Super Taxa
Microorganisms
Organism Name
bacteria
Taxa Notes
Bacteria, Eubacteria, Microorganisms
RN 134736-36-0 (poly(3-hydroxy-5-phenylvalerate))

FULL SEARCH HISTORY

=> d his nofile

(FILE 'HOME' ENTERED AT 10:21:20 ON 30 AUG 2007)

FILE HCAPLUS' ENTERED AT 10:21:28 ON 30 AUG 2007 E US20060211100/PN

Ll I SEA ABB=ON PLU=ON US20060211100/PN D ALL SEL RN

FILE 'REGISTRY' ENTERED AT 10:22:32 ON 30 AUG 2007 56 SEA ABB=ON PLU=ON (686753-12-8/BLOR 10028-15-6/BL OR 112-05-0/BI OR 112-38-9/BI OR 113-24-6/BI OR 119003-51-9/BI OR 121739-61-5/BI OR 13907-47-6/BI OR 14333-13-2/BI OR 147867-05-8/BI OR 15056-35-6/BI OR 153744-07-1/BI OR 16177-21-2/BI OR 173395-00-1/BI OR 198274-26-9/BI OR 20492-10-8/BI OR 21010-06-0/BI OR 2270-20-4/BI OR 27588-56-3/BI OR 325698-86-0/BI OR 358657-18-8/BI OR 358718-36-2/BI OR 3903-40-0/BI OR 404867-92-1/BI OR 4144-62-1/BI OR 4441-63-8/BI OR 446881-43-2/BI OR 452081-78-6/BI OR 452081-80-0/BI OR 454458-22-1/BI OR 457655-22-0/BI OR 463301-93-1/BI OR 477345-08-7/BI OR 483343-33-5/BI OR 484040-48-4/BI OR 50-99-7/BI OR 56721-43-8/BI OR 591251-75-1/BI OR 5962-88-9/BI OR 6303-58-8/BI OR 64740-39-2/BI OR 647831-62-7/BI OR 647831-63-8/BI OR 647831-64-9/BI OR 647831-65-0/BI OR 686753-13-9/BI OR 686753-17-3/BI OR 686753-18-4/BI OR 686753-19-5/BI OR 689291-81-4/BI OR 694533-29-4/BLOR 694533-30-7/BLOR 694533-31-8/BLOR 7170-40-3/BI OR 7722-64-7/BI OR 818-88-2/BI) D SCAN

FILE 'STNGUIDE' ENTERED AT 10:23:23 ON 30 AUG 2007

FILE 'REGISTRY' ENTERED AT 10:29:02 ON 30 AUG 2007 28 SEA ABB=ON PLU=ON L2 AND C6/ES D SCAN

FILE 'STNGUIDE' ENTERED AT 10:31:12 ON 30 AUG 2007

FILE 'LREGISTRY' ENTERED AT 10:34:01 ON 30 AUG 2007 1.4 STR

FILE 'REGISTRY' ENTERED AT 10:38:33 ON 30 AUG 2007

E A/PCT

1.5 I SEA SSS SAM L4 D SCAN

1.6 1299 SEA SSS FUL L4

SAV TEMP L6 LIL689REG/A

10 SEA ABB=ON PLU=ON L2 AND L6 L7 D SCAN

18 134 SEA ABB=ON PLU=ON L6 AND PMS/CI

L9 1165 SEA ABB=ON PLU=ON L6 NOT L8

20 SEA ABB=ON PLU=ON L6 AND (MEDĖINE/LC OR BIOSIS/LC OR L10 DRUGU/LC OR EMBASE/LC) D SCAN

FILE 'HCAPLUS' ENTERED AT 10:46:04 ON 30 AUG 2007

LH 25 SEA ABB=ON PLU=ON L7

949 SEA ABB=ON PLU=ON L6 L12

L13 1 SEA ABB=ON PLU=ON L1 AND L12 D SCAN E PSEUDOMONAS/CT E E3+ALL

L14 57652 SEA ABB=ON PLU=ON PSEUDOMONAS+PFT,OLD,NT/CT

L15 91 SEA ABB=ON PLU=ON L12 AND L14

QUE ABB=ON PLU=ON PSEUDOMONAS? 1.16

L17 QUE ABB=ON PLU=ON POLYHYDROXYALKANOATE OR POLY(W)HYDR OXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOATE

L18 QUE ABB=ON PLU=ON COPOLYM? OR CO(W)POLYM?

L19 QUE ABB=ON PLU=ON L17(3A)L18

- 10/531,689 L.20 99 SEA ABB=ON PLU=ON L12 AND L16 99 SEA ABB=ON PLU=ON L15 OR L20 L21 L22 2313 SEA ABB=ON PLU=ON POLYHYDROXYALKANOATE OR POLY(W)HYDR OXYALKANOATE OR POLY(W)HYDROXY(W)ALKANOATE L23 78 SEA ABB=ON PLU=ON L21 AND L22 L24 8 SEA ABB=ON PLU=ON 1.21 AND L19 D SCAN L25 I SEA ABB=ON PLU=ON L1 AND L24 D L24 1-8 AU **DEL SEL** SEL LI AU 1..26 763 SEA ABB=ON PLU=ON ("FUKUI, TATSUKI"/AU OR "HONMA, TSUTOMU"/AU OR "IMAMURA, TAKESHI"/AU OR "KENMOKU, TAKASHI"/AU OR "KOZAKI, SHINYA"/AU OR "MIHARA, CHIEKO"/AU OR "YANO, TETSUYA"/AU) E SUGAWA E/AU L27 QUE ABB=ON PLU=ON SUGAWA E?/AU L28 786 SEA ABB=ON PLU=ON L27 OR L26 FILE 'ZCAPLUS' ENTERED AT 10:59:47 ON 30 AUG 2007 1.29 QUE ABB=ON PLU=ON FUKULT?/AU L30 QUE ABB=ON PLU=ON HOMA T?/AU QUE ABB=ON PLU=ON IMAMURA T?/AU L31 1.32 QUE ABB=ON PLU=ON KENMOKU T?/AU L33 QUE ABB=ON PLU=ON KOZAKI S?/AU L34 QUE ABB=ON PLU=ON MIHARA C?/AU QUE ABB=ON PLU=ON YANO T?/AU L35 L36 QUE ABB=ON PLU=ON L27 OR (L29 OR L30 OR L31 OR L32 OR L33 OR L34 OR L35) QUE ABB=ON PLU=ON L27 AND L29 AND L30 AND L31 AND L37 L32 AND L33 AND L34 AND L35 FILE HCAPLUS! ENTERED AT 11:03:31 ON 30 AUG 2007 D L1 PA FILE 'ZCAPLUS' ENTERED AT 11:03:31 ON 30 AUG 2007 E CANON/PA E CANON KABUSHIKI/PA L38 QUE ABB=ON PLU=ON (CANON(W)KABUSH!KI?)/PA,CS,SO,CO FILE 'HCAPLUS' ENTERED AT 11:04:52 ON 30 AUG 2007 L39 0 SEÁ ABB=ON PLU=ON L27 AND L29 AND L30 AND L31 AND L32 AND L33 AND L34 AND L35 L40 O SEA ABB=ON PLU=ON L27 AND L29 AND L31 AND L32 AND L33 AND L34 AND L35 1.41 9410 SEA ABB=ON PLU=ON L27 OR (L29 OR L30 OR L31 OR L32 OR L33 OR L34 OR L35) 122 SEA ABB=ON PLU=ON L41 AND L38 1.42 L43 119 SEA ABB=ON PLU=ON (L42 OR L28) AND L16 L44 80 SEA ABB=ON PLU=ON L43 AND L17 1.45 21 SEA ABB=ON PLU=ON L43 AND L18 D I-3 KWIC QUE ABB=ON PLU=ON PY<2003 OR PRY<2003 OR AY<2003 OR L46 MY<2003 OR REVIEW/DT L47 20 SEA ABB=ON PLU=ON L45 AND L46 SAV L47 L1L689HCPIN/A L48 24 SEA ABB=ON PLU=ON L23 AND L18 149 24 SEA ABB=ON PLU=ON L48 OR L24 1.50 21 SEA ABB=ON PLU=ON L49 AND L46 1.51 8 SEA ABB=ON PLU=ON L50 AND L24 1.52 11 SEA ABB=ON PLU=ON L50 NOT L47 SAV L52 LIL689HCP/A D QUE L52 FILE 'MEDLINE, BIOSIS, DRUGU, EMBASE' ENTERED AT 11:14:08 ON 30 AUG 2007 1.53 67 SEA ABB=ON PLU=ON L10 L54 3 SEA ABB=ON PLU=ON L53 AND L16
- 1.55 2 SEA ABB=ON PLU=ON L53 AND L17 L56 4 SEA ABB=ON PLU=ON L54 OR L55 L57 255 SEA ABB=ON PLU=ON L28 L58 10077 SEA ABB=ON PLU=ON L36 OR L57 60 SEA ABB=ON, PLU=ON L58 AND L38 .1.59 L60 34 SEA ABB=ON PLU=ON L59 AND (L16 OR L17)

MC

56-56

L61 7 SEA ABB=ON PLU=ON L60 AND L46

D SCAN

SAV L61 LIL689MULTIN/A

1.62 4 SEA ABB=ON PLU=ON L56 NOT L61

SAV L62 LIL689MULT/A

FILE 'STNGUIDE' ENTERED AT 11:19:54 ON 30 AUG 2007

D QUE LA7 D QUE L61

L63

FILE 'HCAPLUS, BIOSIS' ENTERED AT 11:22:00 ON 30 AUG 2007

27 DUP REM L47 L61 (0 DUPLICATES REMOVED)

ANSWERS '1-20' FROM FILE HCAPLUS ANSWERS '21-27' FROM FILE BIOSIS

D L63 1-27 IBIB ED AB

D QUE STAT L52

D QUE STAT L62

FILE 'HCAPLUS, MEDLINE, BIOSIS' ENTERED AT 11:23:35 ON 30 AUG 2007

L64 15 DUP REM L52 L62 (0 DUPLICATES REMOVED)

ANSWERS '1-11' FROM FILE HCAPLUS

ANSWER '12' FROM FILE MEDLINE

ANSWERS '13-15' FROM FILE BIOSIS

D L64 1-11 IBIB ED ABS HITSTR HITIND

D L64 12-15 IBIB ED AB HIT IND